

Synthesis of Amine- and Thiol-Modified Nucleoside Phosphoramidites for Site-Specific Introduction of Biophysical Probes into RNA

Shengxi Jin,† Chandrasekhar V. Miduturu,† David C. McKinney, and Scott K. Silverman* Department of Chemistry, University of Illinois, 600 South Mathews Avenue, Urbana, Illinois 61801

scott@scs.uiuc.edu

Received January 11, 2005

Phosphoramidites for solid-phase RNA synthesis

For studies of RNA structure, folding, and catalysis, site-specific modifications are typically introduced by solid-phase synthesis of RNA oligonucleotides using nucleoside phosphoramidites. Here, we report the preparation of two complete series of RNA nucleoside phosphoramidites; each has an appropriately protected amine or thiol functional group. The first series includes each of the four common RNA nucleotides, U, C, A, and G, with a 2'-(2-aminoethoxy)-2'-deoxy substitution (i.e., a primary amino group tethered to the 2'-oxygen by a two-carbon linker). The second series encompasses the four common RNA nucleotides, each with the analogous 2'-(2-mercaptoethoxy)-2'-deoxy substitution (i.e., a tethered 2'-thiol). The amines are useful for acylation and reductive amination reactions, and the thiols participate in displacement and oxidative cross-linking reactions, among other likely applications. The new phosphoramidites will be particularly valuable for enabling site-specific introduction of biophysical probes and constraints into RNA.

Introduction

Appending chromophores and other biophysical probes and constraints onto biological macromolecules enables us to develop a detailed understanding of their structure, folding, and catalysis. For such purposes, an important prerequisite is the development of the chemical methods necessary to synthesize the modified biomolecules. Here, we describe the preparation of eight modified RNA nucleoside phosphoramidites. These reagents collectively allow the solid-phase synthesis of RNA oligonucleotides that incorporate a primary amine or thiol group tethered

S.J. and C.V.M. are co-first authors of this paper. (1) (a) Cohen, S. B.; Cech, T. R. J. Am. Chem. Soc. 1997, 119, 6259-6268. (b) Glick, G. D. Biopolymers 1998, 48, 83–96. (c) Sigurdsson, S. T. Methods 1999, 18, 71–77. (d) Silverman, S. K.; Cech, T. R. Biochemistry 1999, 38, 14224–14237. (e) Silverman, S. K.; Deras, M. L.; Woodson, S. A.; Scaringe, S. A.; Cech, T. R. Biochemistry 2000, 39,

12465–12475. (f) Silverman, S. K.; Cech, T. R. RNA **2001**, 7, 161–166. (g) Young, B. T.; Silverman, S. K. Biochemistry **2002**, 41, 12271– 12276. (h) Smalley, M. K.; Silverman, S. K. 2005, submitted for publication.

CHART 1. RNAs with Site-Specific 2'-Tethered Amine or 2'-Tethered Thiol Groups

2'-(2-aminoethoxy)-2'-deoxy-RNA "tethered amine"

2'-(2-mercaptoethoxy)-2'-deoxy-RNA "tethered thiol"

to the ribose 2'-oxygen of any specific nucleotide via a two-carbon tether (Chart 1). The ribose 2'-position is particularly useful for chemical modification because of its location in the minor groove of A-form duplex nucleic acid, where it points outward into solution.2 The ribose 2'-position is also readily modified synthetically, as described in this work. Both established³ and emerging⁴

^{*} To whom correspondence should be addressed. Phone: (217) 244-4489. Fax: (217) 244-8024.

⁽²⁾ Saenger, W. Principles of Nucleic Acid Structure; Springer-Verlag: New York, 1984.

ligation techniques are used to construct RNAs larger than those that can be prepared by direct solid-phase synthesis (<100 nt),⁵ as required for the study of many catalytically active RNAs. 6 In conjunction with such techniques, the modified nucleoside phosphoramidites reported here should be immediately useful in ongoing biochemical and biophysical studies of nucleic acids.

We chose to prepare phosphoramidites with relatively short two-carbon tethers rather than longer tethers for two specific reasons. First, for some purposes, such as certain fluorescence applications, a long tether can provide too little interaction between the RNA and a chromophore appended onto the tether. Because a spacer of variable length can typically be incorporated during the reaction with a chromophore reagent, the most generally useful tethers for such applications are relatively short. Second, for applications involving structural constraints, a long tether may introduce too much flexibility into the RNA, thereby precluding the goal of a structural constraint. In contrast, a short tether permits more structural control over the RNA conformation. For such applications, the two-carbon tethers in Chart 1 are the shortest chemically stable tethers that retain the 2'oxygen atom (thereby avoiding certain concerns over destabilizing RNA structure with unnatural functional groups⁷) and also provide reactive amine or thiol functional groups for subsequent derivatization. Our synthetic routes to the nucleoside phosphoramidites that are required to incorporate these tethers into RNA are designed to be maximally convergent. Experimental protocols for preparing sufficient quantities of each phosphoramidite (at least 100 mg) for conventional solidphase RNA oligonucleotide synthesis are described.

Results and Discussion

Pyrimidine Nucleoside Phosphoramidites with 2'-Tethered Amines. A small number of pyrimidine nucleosides with a primary amino group tethered via the ribose 2'-oxygen have been reported previously. For tethers longer than two carbon atoms, the only reported

approach has been the reaction of a pyrimidine nucleoside 2',3'-stannylene derivative⁸ with an alkyl halide. The reaction proceeds with a modest chemical yield and little, if any, selectivity between 2'- and 3'-alkylation,9 in those cases for which the yield and selectivity were reported. 10 We chose to avoid such alkylation routes to prevent separating mixtures of 2'- and 3'-isomers. For the twocarbon tethers, others have reported nucleophilic attack on a 2,2'-anhydrothymidine derivative, either by a boron derivative of 2-hydroxyethylphthalimide¹¹ or by the latter reagent itself in the presence of $Ti(OiPr)_4$. The former procedure requires a sealed stainless steel bomb heated to 150 °C and provides only a 21% yield of 2'-modified product after silica gel column chromatography; the yield is 18% for a three-carbon tether and anhydrouridine. 13 The latter procedure with Ti(OiPr)₄ is reported to produce a 32% yield, but no experimental protocol was provided. An alternative approach to the 2'-modified pyrimidine nucleosides first protects the 3'- and 5'-hydroxyl groups as a TIPDS silvl derivative (see below); this step is followed by the alkylation of the 2'-hydroxyl group with an α -bromo ester. 14,15 Then, reduction of the ester produces the 2'-OCH2CH2OH substitutent for further manipulation. In one case, the tethered hydroxyl group was activated with TsCl and displaced with NaN₃ to introduce the nitrogen, but the communication provided no experimental protocols. 14 In another case, phthalimide was used in a Mitsunobu reaction to introduce the nitrogen, and again no experimental protocol was provided.¹⁵

On the basis of these limited precedents, we decided to protect the pyrimidine 3'- and 5'-hydroxyl groups at the outset and to employ a different 2'-alkylation strategy. Our route began with the simultaneous TIPDS (tetraisopropyldisiloxane-1,3-diyl) protection¹⁶ of the 3'and 5'-hydroxyl groups of uridine **U1** (Scheme 1).¹⁷ The TIPDSCl₂ silylating reagent is commercially available, albeit expensive. As an alternative, TIPDSCl₂ may conveniently be generated from the corresponding bissilane and CCl₄ using substoichiometric PdCl₂. ¹⁸ The crude TIPDS-uridine was immediately protected as its

^{(3) (}a) Moore, M. J.; Sharp, P. A. Science 1992, 256, 992-997. (b) Moore, M. J.; Query, C. C. In RNA-Protein Interactions: A Practical Approach; Smith, C. W. J., Ed.; Oxford University Press: Oxford, U.K., 1998; pp 75–108. (c) Moore, M. J. Methods Mol. Biol. 1999, 118, 11– 19. (d) Moore, M. J.; Query, C. C. Methods Enzymol. 2000, 317, 109-

^{(4) (}a) Silverman, S. K. Org. Biomol. Chem. 2004, 2, 2701-2706. (4) (a) Silverman, S. K. Org. Biomot. Chem. 2004, 2, 2101–2106. (b) Flynn-Charlebois, A.; Wang, Y.; Prior, T. K.; Rashid, I.; Hoadley, K. A.; Coppins, R. L.; Wolf, A. C.; Silverman, S. K. J. Am. Chem. Soc. 2003, 125, 2444–2454. (c) Flynn-Charlebois, A.; Prior, T. K.; Hoadley, K. A.; Silverman, S. K. J. Am. Chem. Soc. 2003, 125, 5346–5350. (d) Wang, Y. Silverman, S. K. J. Am. Chem. Soc. 2003, 125, 5369–6381 Wang, Y.; Silverman, S. K. J. Am. Chem. Soc. 2003, 125, 6880-6881. (e) Wang, Y.; Silverman, S. K. Biochemistry 2003, 42, 15252-15263.
 (f) Ricca, B. L.; Wolf, A. C.; Silverman, S. K. J. Mol. Biol. 2003, 330, 1015–1025. (g) Prior, T. K.; Semlow, D. R.; Flynn-Charlebois, A.; Rashid, I.; Silverman, S. K. Nucleic Acids Res. 2004, 32, 1075–1082. (h) Coppins, R. L.; Silverman, S. K. Nat. Struct. Mol. Biol. 2004, 11, 270-274. (i) Coppins, R. L.; Silverman, S. K. J. Am. Chem. Soc. 2004, 126, 16426-16432. (j) Coppins, R. L.; Silverman, S. K. J. Am. Chem. Soc. 2005, 127, 2900–2907. (k) Wang, Y.; Silverman, S. K. Biochemistry 2005, 44, 3017–3023. (1) Semlow, D. R.; Silverman, S. K. J. Mol. Evol. **2005**, in press

⁽⁵⁾ Marshall, W. S.; Kaiser, R. J. Curr. Opin. Chem. Biol. 2004, 8,

^{(6) (}a) Cech, T. R. Annu. Rev. Biochem. 1990, 59, 543-568. (b) Michel, F.; Ferat, J. L. Annu. Rev. Biochem. 1995, 64, 435-461. (c) Frank, D. N.; Pace, N. R. Annu. Rev. Biochem. 1998, 67, 153-180.

⁽⁷⁾ Pham, J. W.; Radhakrishnan, I.; Sontheimer, E. J. Nucleic Acids Res. 2004, 32, 3446-3455.

⁽⁸⁾ Wagner, D.; Verheyden, J. P. H.; Moffatt, J. G. J. Org. Chem. **1974**, 39, 24-30.

⁽⁹⁾ Manoharan, M.; Tivel, K. L.; Andrade, L. K.; Cook, P. D. Tetrahedron Lett. 1995, 36, 3647-3650.

⁽¹⁰⁾ Griffey, R. H.; Monia, B. P.; Cummins, L. L.; Freier, S.; Greig, M. J.; Guinosso, C. J.; Lesnik, E.; Manalili, S. M.; Mohan, V.; Owens, S.; Ross, B. R.; Sasmor, H.; Wancewicz, E.; Weiler, K.; Wheeler, P. D.; Cook, P. D. *J. Med. Chem.* **1996**, *39*, 5100–5109.

Соок, Р. D. J. Med. Chem. 1996, 39, 5100—5109. (11) (a) Manoharan, M.; Prakash, T. P.; Barber-Peoc'h, I.; Bhat, B.; Vasquez, G.; Ross, B. S.; Cook, P. D. J. Org. Chem. 1999, 64, 6468—6472. (b) Prakash, T. P.; Puschl, A.; Lesnik, E.; Mohan, V.; Tereshko, V.; Egli, M.; Manoharan, M. Org. Lett. 2004, 6, 1971—1974. (12) Blommers, M. J.; Natt, F.; Jahnke, W.; Cuenoud, B. Biochemistry 1998, 37 17714—17795

istry 1998, 37, 17714-17725.

⁽¹³⁾ Ross, B. S.; Springer, R. H.; Tortorici, Z.; Dimock, S. *Nucleosides Nucleotides* **1997**, *16*, 1641–1643. (14) Cuenoud, B.; Casset, F.; Hüsken, D.; Natt, F.; Wolf, R. M.;

Altmann, K.-H.; Martin, P.; Moser, H. E. Angew. Chem., Int. Ed. 1998, 37, 1288-1291.

^{(15) (}a) Sollogoub, M.; Dominguez, B.; Fox, K. R.; Brown, T. Chem. Commun. 2002, 2315-2316. (b) Osborne, S. D.; Powers, V. E.; Rusling, D. A.; Lack, O.; Fox, K. R.; Brown, T. Nucleic Acids Res. 2004, 32, 4439 - 4447

^{(16) (}a) Markiewicz, W. T.; Markiewicz, M. Nucleic Acids Res., Spec. Publ. 1978, 4, 185. (b) Markiewicz, W. T. J. Chem. Res., Synop. 1979, 24–25. (c) Markiewicz, W. T. J. Chem. Res., Miniprint 1979, 181–197. (d) Markiewicz, W. T.; Biala, E.; Adamiak, R. W.; Grzeskowiak, K.; Kierzek, R.; Kraszewski, A.; Stawinski, J.; Wiewiorowski, M.

Nucleic Acids Symp. Ser. 1980, 115–127.
(17) Robins, M. J.; Wilson, J. S.; Hansske, F. J. Am. Chem. Soc. **1983**, 105, 4059-4065.

 N^3 -methoxycarbonylvinyl (mocvinyl) derivative, 19 U2; 20 this protection was necessary to avoid undesired derivatization of N³ in the subsequent alkylation step. Palladium-catalyzed allylation²¹ of the free 2'-hydroxyl group of U2 readily produced U3 in high yield. This 2'-O-allyl nucleoside was ozonolyzed and treated with NaBH4 to provide 2'-(2-hydroxyethyl) nucleoside **U4**. The ozonolysis conditions also removed the mocvinyl group, presumably by the oxidative cleavage of the C=C double bond and the subsequent breakdown of the hemiaminal. TIPDSprotected nucleoside U4 has been accessed before by different routes, 22-24 but at most partial experimental protocols were provided; in all cases, our route is one to five steps shorter.

The tethered hydroxyl group of **U4** was exchanged for an azido moiety by mesylation to $U5^{23}$ and displacement with sodium azide in the presence of 18-crown-6, forming **U6**. In a one-pot procedure, the azide of **U6** was reduced via the Staudinger reaction (PPh₃, H₂O) to the amino group, which was protected as the trifluoromethyl amide, and the TIPDS 3',5'-protecting group was removed with TBAF, producing U7.14 The Staudinger reaction was used instead of H₂ reduction over Pd/C because, in our hands, the latter method led to \sim 15% hydrogenation of the uracil C⁵=C⁶ double bond.²⁵ Trifluoromethyl amide protection was chosen because the trifluoromethyl group is readily removed by transamidation during the standard NH₃ deprotection of RNA oligonucleotides following solidphase synthesis.26

Selective reprotection of the 5'-hydroxyl group of U7 as its 4,4'-dimethoxytrityl (DMT) derivative led to U8, which was phosphitylated with 2-cyanoethyl N, N, N', N'tetraisopropylphosphorodiamidite using 4,5-dicyanoimidazole (DCI) as the activator²⁷ to provide uridinetethered amine nucleoside phosphoramidite **U9**. Starting with uridine U1, the overall route provides phosphoramidite **U9** in a 20% overall yield in 9 steps with 7 chromatographic purifications; 440 mg of U9 was obtained. This route to the 2'-(2-aminoethoxy)-substituted pyrimidine phosphoramidite represents a substantial improvement in yield and convenience for moderate-scale laboratory synthesis when compared to previous reports for similar compounds, as cited above.

⁽¹⁸⁾ Ferreri, C.; Costantino, C.; Romeo, R.; Chatgilialoglu, C.

Tetrahedron Lett. 1999, 40, 1197–1200. (19) Faja, M.; Ariza, X.; Gálvez, C.; Vilarrasa, J. Tetrahedron Lett. **1995**, 36, 3261–3264.

⁽²⁰⁾ Costa, A. M.; Faja, M.; Farràs, J.; Vilarrasa, J. Tetrahedron Lett. 1998, 39, 1835-1838.

^{(21) (}a) Lakhmiri, R.; Lhoste, P.; Sinou, D. Tetrahedron Lett. 1989, 30, 4669-4672. (b) Sproat, B. S.; Iribarren, A.; Beijer, B.; Pieles, U.; Lamond, A. I. Nucleosides Nucleotides 1991, 10, 25-36. (c) Kachalova, A. V.; Zatsepin, T. S.; Romanova, E. A.; Stetsenko, D. A.; Gait, M. J.; Oretskaya, T. S. Nucleosides, Nucleotides Nucleic Acids 2000, 19, 1693 - 1707.

⁽²²⁾ Douglas, M. E.; Beijer, B.; Sproat, B. S. Bioorg. Med. Chem. Lett. 1994, 4, 995-1000.

⁽²³⁾ Maglott, E. J.; Glick, G. D. Nucleic Acids Res. 1998, 26, 1301-1308

⁽²⁴⁾ Dobson, N.; McDowell, D. G.; French, D. J.; Brown, L. J.; Mellor, J. M.; Brown, T. Chem. Commun. 2003, 1234-1235.

^{(25) (}a) Watkins, B. E.; Kiely, J. S.; Rapoport, H. J. Am. Chem. Soc. 1982, 104, 5702-5708. (b) Johnson, D. C., II; Widlanski, T. S. Org. Lett. 2004, 6, 4643-4646.

^{(26) (}a) Imazawa, M.; Eckstein, F. J. Org. Chem. 1979, 44, 2039-2041. (b) Pieken, W. A.; Olsen, D. B.; Benseler, F.; Aurup, H.; Eckstein, F. Science 1991, 253, 314-317. (c) Benseler, F.; Williams, D. M.; Eckstein, F. Nucleosides Nucleotides 1992, 11, 1333-1351.

⁽²⁷⁾ Vargeese, C.; Carter, J.; Yegge, J.; Krivjansky, S.; Settle, A.; Kropp, E.; Peterson, K.; Pieken, W. Nucleic Acids Res. 1998, 26, 1046–

U9 was converted to its triazolide derivative U10 to obtain the cytidine phosphoramidite analogue (Scheme 1). This reagent can be incorporated as usual during solid-phase RNA synthesis and upon standard oligonucleotide deprotection with NH₃ leads to cytidine at the desired nucleotide position.²⁸ This highly convergent route is greatly preferred over the independent synthesis of the cytidine nucleoside phosphoramidite starting from cytidine.²⁹ Such an independent route would entail nearly twice the number of reaction steps as synthesis of the uridine nucleoside phosphoramidite U9 plus its one-step conversion to the triazolide U10. Starting from uridine U1, phosphoramidite U10 was obtained in a 10% overall yield in 10 steps with 8 chromatographic purifications; 100 mg of U10 was prepared.

Pyrimidine Nucleoside Phosphoramidites with 2'-Tethered Thiols. The route to the pyrimidine nucleosides with 2'-tethered thiols started with intermediate **U4** from the pyrimidine-tethered amines strategy (Scheme 2). Displacement of the hydroxyl group with trityl mercaptan under Mitsunobu conditions produced tritylprotected thiol U11.22 The conversion of U4 to U11 appears to be the first reported use of trityl mercaptan as the nucleophile in a Mitsunobu reaction. Intermediate U11 was treated with TBAF to remove the 3',5'-protecting group, providing diol U12. As was done for the amine route, the 5'-hydroxyl was reprotected as its DMT derivative to produce **U13**, which was phosphitylated to provide uridine-tethered thiol nucleoside phosphoramidite U14. The overall route from uridine **U1** to **U14** is 8 steps with 7 chromatographic purifications and proceeds in a 13% overall yield; 340 mg of U14 was produced. Compounds U11-U14 were previously accessed by a different route with little experimental information provided;²² our route

to **U11** is three steps shorter and has a higher yield starting from uridine (30% vs 16–20%). The cytidine phosphoramidite analogue was obtained from **U14** as the triazolide derivative **U15**, in a manner analogous to the tethered amine route. A cytidine-tethered thiol was prepared previously by a completely different route that did not involve the uridine triazolide.²⁹ The route from **U1** to **U15** is 9 steps with 8 chromatographic purifications in 7% overall yield; 112 mg of **U15** was prepared.

The trityl group has been used previously for thiol protection of nucleosides. 30,31 For the 2'-tethered thiol, we chose trityl rather than tert-butyl disulfide protection 23 because the S-trityl deprotection procedure (treatment with $AgNO_3^{31}$) avoids the use of reducing agents such as DTT. Application of such reducing agents would complicate the subsequent use of oligonucleotides with tert-butyl disulfide protecting groups in RNA ligation reactions catalyzed by protein enzymes, which are normally used in DTT-containing buffers. 3

Adenosine Nucleoside Phosphoramidite with 2'-Tethered Amine. Our preparation of the adenosine nucleoside phosphoramidite with a 2'-tethered amine followed a route related to that used for the pyrimidines (Scheme 3). Adenosine A1 was alkylated preferentially at the 2'-position^{10,32} with NaH and methyl bromoacetate to provide A2 in a 47% yield.³³ Although ethyl bromoacetate had a slightly higher yield of 58%, greater than 50% of the ethyl ester analogue of A2 underwent transesterification during silica gel chromatography with MeOH/CH₂Cl₂ as the eluent (for separation purposes, methanol proved to be superior to ethanol as the chromatography cosolvent). Therefore, although the yield was

^{(28) (}a) Reese, C. B.; Skone, P. A. J. Chem. Soc., Perkin Trans. 1 1984, 1263–1271. (b) Xu, Y.-Z.; Zheng, Q.; Swann, P. F. J. Org. Chem. 1992, 57, 3839–3845. (c) Shah, K.; Wu, H.; Rana, T. M. Bioconjugate Chem. 1994, 5, 508–512. (d) Obika, S.; Uneda, T.; Sugimoto, T.; Nanbu, D.; Minami, T.; Doi, T.; Imanishi, T. Bioorg. Med. Chem. Lett. 2001, 9, 1001–1011. (e) Chirakul, P.; Sigurdsson, S. T. Org. Lett. 2003, 5, 917–

⁽²⁹⁾ Goodwin, J. T.; Osborne, S. E.; Scholle, E. J.; Glick, G. D. J. Am. Chem. Soc. **1996**, 118, 5207–5215.

^{(30) (}a) Connolly, B. A.; Rider, P. *Nucleic Acids Res.* **1985**, *13*, 4485–4502. (b) Manoharan, M.; Tivel, K. L.; Ross, B.; Cook, P. D. *Gene* **1994**, *149*, 147–156.

⁽³¹⁾ Hamm, M. L.; Piccirilli, J. A. J. Org. Chem. 1997, 62, 3415-3420

^{(32) (}a) Manoharan, M.; Guinosso, C. J.; Cook, P. D. Tetrahedron Lett. **1991**, 32, 7171–7174. (b) Manoharan, M.; Johnson, L. K.; Tivel, K. L.; Springer, R. H.; Cook, P. D. Bioorg. Med. Chem. Lett. **1993**, 3, 2765–2770.

⁽³³⁾ Kawasaki, A. M.; Casper, M. D.; Prakash, T. P.; Manalili, S.; Sasmor, H.; Manoharan, M.; Cook, P. D. *Tetrahedron Lett.* **1999**, 40, 661–664.



slightly lower, we chose to use methyl bromoacetate in the alkylation reaction so that the product, **A2**, was homogeneous and fully characterizable by 2D NMR spectroscopy. In the 2D HMBC spectrum of **A2**, both of the long-range (four-bond) H-C-O-C couplings between H^2 and $2'-OCH_2$ and between $2'-OCH_2$ and C^2 were clearly observed, thereby confirming the identity of **A2** as the 2'-alkylated isomer.

Nucleoside **A2** was doubly protected with *tert*-butyl-dimethylsilyl (TBS) at its 3′- and 5′-positions to produce **A3**. We used double TBS protection rather than 3′,5′-TIPDS protection because the site-selectivity of TIPDSCl₂ for the 3′- and 5′-positions was not required (the 2′-hydroxyl was already substituted) and TBSCl is less expensive than TIPDSCl₂. Nucleoside **A3** was reduced with LiAlH₄ to provide the 2′-(2-hydroxyethyl) nucleoside **A4**. In our hands, NaBH₄ was ineffective for reductions of this type. 33

Our overall strategy to the 2'-(2-hydroxyethyl)adenosine nucleoside is an alternative to the previously used 2'-O-allylation of adenosine followed by ozonolysis and reduction of the tethered aldehyde group, which was performed with a different combination of hydroxyl and nucleobase protecting groups. ³⁴ Here, we favored adenosine 2'-alkylation with the α -bromo ester because we found that the 2'-isomer, **A2**, and its 3'-isomer could easily be separated by silica gel chromatography, which was not the case for the 2'-O-allyl derivatives. ³⁴ We also

Protection of nucleobase N⁶ of **A4** as its isobutyryl amide was achieved with transient TMS protection³⁵ of the 2'-tethered hydroxyl group, leading to **A5**. The 2'-tethered hydroxyl group was exchanged for an azido group by mesylation to **A6**, which was followed by azide displacement yielding **A7**. The amine was obtained by reduction with H₂ over Pd/C and then protected as trifluoromethyl amide **A8**. Removal of the 3'- and 5'-silyl groups with TBAF, yielding **A9**, was followed by 5'-DMT protection to produce **A10** and phosphitylation to yield adenosine tethered amine nucleoside phosphoramidite **A11**. The route from adenosine **A1** to **A11** takes 10 steps with 8 chromatographic purifications in a 12% overall yield; 198 mg of phosphoramidite **A11** was prepared.

Adenosine Nucleoside Phosphoramidite with 2'-Tethered Thiol. Adenosine nucleoside derivatives with 2'-OCH₂CH₂SR substituents appear to be unreported. The adenosine-tethered thiol phosphoramidite was prepared from intermediate A5 by a route analogous to that used for the pyrimidine thiol phosphoramidites (Scheme 4). Mitsunobu displacement with trityl mercaptan produced A12 from A5; A12 was 3',5'-deprotected to form

found that direct 2'-alkylation of adenosine with functionalized alkyl halides such as N-(2-bromoethyl)phthalimide, which was at least moderately successful with longer-chain substituted alkyl halides, 9,10,32 was not preparatively useful for the two-carbon tether (R. L. Coppins and S. K. Silverman, data not shown).

⁽³⁴⁾ Prakash, T. P.; Kawasaki, A. M.; Fraser, A. S.; Vasquez, G.; Manoharan, M. J. Org. Chem. **2002**, 67, 357–369.

⁽³⁵⁾ Ti, G. S.; Gaffney, B. L.; Jones, R. A. J. Am. Chem. Soc. 1982, 104, 1316–1319.

A13, and then A13 was 5'-DMT-protected to produce A14. Finally, phosphitylation yielded the adenosine-tethered thiol nucleoside phosphoramidite A15. Phosphoramidite A15 was obtained in 8 steps with 7 chromatographic purifications from adenosine A1 in a 12% overall yield; 102 mg of A15 was synthesized.

Guanosine Nucleoside Phosphoramidite with 2'-**Tethered Amine.** Direct 2'-alkylation of unprotected guanosine is impractical because of the unwanted reactivity of the nucleobase. 36,37 Therefore, our route to the guanosine derivatives began with commercially available 2,6-diaminopurine ribonucleoside G1 (Scheme 5). Nucleoside G1 was preferentially 2'-alkylated with NaH and benzyl 2-bromoethyl ether to provide G2, which was obtained free of the 3'-isomer after silica gel chromatography.³⁸ Although the yield of **G2** was modest (33%), the reaction and separation were readily performed on a large scale, and for expediency we judged this route preferable to multistep options involving protecting groups (e.g., initial 3',5'-TIPDS protection of G1, then 2'alkylation and 3',5'-deprotection). The enzymatic conversion of the 2,6-diaminopurine derivative G2 to the guanosine analogue with adenosine deaminase (ADA) provided 2'-functionalized guanosine nucleoside G3.38,39 Protection of the 3'- and 5'-hydroxyl groups using 5 equiv of TBSCl was accompanied by guanine O^6 -silvlation to form **G4**. This compound appears to be stable to silica gel chromatography, which is somewhat unexpected on the basis of literature reports for similar compounds.³⁷ Isobutyryl protection of the guanosine N² was achieved along with removal of the O^6 -TBS group, producing **G5**. The 2'-tethered benzyl group was hydrogenolyzed to provide G6, which was converted via the mesylate G7 to the azide G8. Reduction with H_2 over Pd/C to the amine, protection as the trifluoromethyl amide, and removal of the 3'- and 5'-silyl groups with TBAF produced G9. The 5'-hydroxyl was protected as its DMT derivative, yielding G10. To complete the route, the nucleoside was phosphitylated to produce the guanosine-tethered amine nucleoside phosphoramidite G11. The 10-step route from 2,6-diaminopurine ribonucleoside G1 to G11 was achieved with 8 chromatographic purifications in a 6% overall yield. This includes the modest 33% yield for the first-step conversion of G1 into the 2'-alkylated G2. A total of 160 mg of G11 was synthesized.

Guanosine Nucleoside Phosphoramidite with 2'-Tethered Thiol. The last of the eight phosphoramidites, the guanosine-tethered thiol, was prepared starting from intermediate G6 (Scheme 6). Exchange of the 2'-tethered hydroxyl group for the trityl-protected thiol group was achieved under Mitsunobu conditions, and the 3'- and 5'-silyl groups were removed with TBAF to produce G12. 5'-DMT protection led to G13, and phosphitylation provided the guanosine-tethered thiol nucleoside phosphoramidite G14. The overall 8-step route from G1 to G14 proceeded with 7 chromatographic purifications in a 7% overall yield; 109 mg of G14 was prepared.

Solid-Phase RNA Synthesis Using the Modified Phosphoramidites. The new nucleoside phosphoramidites were successfully incorporated into RNA oligonucleotides by solid-phase synthesis. For the tethered amine series, each of the four nucleoside phosphoramidites, U14, U15, A11, and G11, was incorporated into a separate oligonucleotide, as detailed in the Experimental Section. This demonstrates that all nucleobase deprotections and the U-triazolide-to-C conversion were successful in the context of oligonucleotides. For the tethered thiol series, adenosine phosphoramidite A15 was used for solid-phase synthesis, thereby establishing that the thiol modification is successfully incorporated. Subsequently, in a larger RNA to which the tethered thiol oligonucleotide derived from A15 was ligated, 40 a standard AgNO₃ deprotection procedure³¹ was employed, which demon-

⁽³⁶⁾ Wagner, E.; Oberhauser, B.; Holzner, A.; Brunar, H.; Issakides, G.; Schaffner, G.; Cotten, M.; Knollmuller, M.; Noe, C. R. *Nucleic Acids Res.* **1991**, *19*, 5965–5971.

^{(37) (}a) Grøtli, M.; Douglas, M. E.; Beijer, B.; Sproat, B. *Bioorg. Med. Chem. Lett.* **1997**, 7, 425–428. (b) Grøtli, M.; Douglas, M.; Beijer, B.; García, R. G.; Eritja, R.; Sproat, B. *J. Chem. Soc., Perkin Trans.* 1 **1997**, 2779–2788

 $^{(38) \} Gundlach, \ C. \ W., \ IV; \ Ryder, \ T. \ R.; \ Glick, \ G. \ D. \ Tetrahedron \ Lett. \ 1997, 38, 4039-4042.$

⁽³⁹⁾ Robins, M. J.; Zou, R.; Hansske, F.; Wnuk, S. F. Can. J. Chem. 1997, 75, 762–767.

Jin et al.

SCHEME 5

SCHEME 6

strates that the free thiol moiety can be readily unmasked after solid-phase synthesis.

For practical purposes, we generally store large amounts of all nucleosides as the immediate precursors to the phosphoramidites (i.e., as U8, U13, A10, A14, G10, and G13), which obviates the issue of long-term phosphoramidite stability. Nevertheless, we have not noticed substantial decomposition of any of the phosphoramidites

(40) Silverman, S. K.; Cech, T. R. Biochemistry 1999, 38, 8691-8702.

upon storage for months at -20 °C or below when the compounds are suitably protected from moisture.

Conclusions

We have described a comprehensive and streamlined set of experimental procedures for synthesizing all four standard RNA nucleoside phosphoramidites, each with either a 2'-tethered amine or a 2'-tethered thiol functional group. The synthetic routes are maximally convergent and provide sufficient amounts of phosphoramidite for several solid-phase synthesis coupling reactions. The resulting oligonucleotides are immediately useful in ongoing studies of RNA structure, folding, and catalysis.

Experimental Section

3',5'-O-(Tetraisopropyldisiloxane-1,3-diyl)-N³-(methoxycarbonylvinyl)uridine (U2). TIPDSCl₂ (2.88 mL, 9.01 mmol) was added dropwise over 5 min to a solution of uridine U1 (2.00 g, 8.20 mmol) in pyridine (50 mL) at 0 °C. (Alternatively, the TIPDSCl₂ can be generated in situ as described in the text; we have successfully used the procedure of ref 18 on the 10-25 mmol scale.) The colorless solution was warmed to room temperature, stirred for 10 h, and then quenched with water (5 mL). Pyridine was removed under vacuum, and the residual white solid was partitioned between EtOAc (100 mL) and water (100 mL). The organic layer was washed with cold 1 N HCl (2 \times 50 mL), saturated aqueous NaHCO₃ (2 \times 50 mL), and saturated aqueous NaCl (50 mL). The organic layer was dried over Na₂SO₄ and concentrated under vacuum to produce 4.20 g of a crude white solid that was used without further purification. This solid was dissolved in 42 mL of 1:5 CH₂Cl₂/CH₃CN and cooled to ~13−15 °C (chilled water bath; at higher temperature, methyl propiolate reacts considerably with the 2'-hydroxyl group, whereas at lower temperature, the overall reaction rate is very low). DMAP (0.30 g, 2.40 mmol) was added to the solution; this was followed by the addition of methyl propiolate (0.25 mL, 2.40 mmol) dropwise over a period of 2 min. An additional aliquot of methyl propiolate (2.40 mmol) was added every 30 min until TLC indicated complete consumption of 3',5'-O-(tetraisopropyldisiloxane-1,3diyl)uridine; three additional aliquots were required. The solution was quenched with MeOH (8 mL) and concentrated under vacuum. The residual red viscous oil was purified via chromatography using 0-30% EtOAc in hexanes as the eluant to yield 3.20 g (68%) of **U2** as a white foam: R_f 0.28 (3:1 hexanes/EtOAc); ¹H NMR (500 MHz, CDCl₃) δ 8.27 (d, J =15.0 Hz, 1H), 7.71 (d, J = 8.0 Hz, 1H), 7.08 (d, J = 15.0 Hz, 1H), 5.76 (d, J = 8.0 Hz, 1H), 5.74 (br s, 1H), 4.37 (dd, J =9.0, 4.7 Hz, 1H), 4.22 (ABqd, J = 13.3, 1.5 Hz, 1H), 4.17 (d, J = 5.0 Hz, 1H, 4.14 - 4.09 (m, 1H), 4.00 (ABqd, J = 13.5, 3.0 (ABqd, J =Hz, 1H), 3.78 (s, 3H), 2.84 (s, 1H), 1.10-1.00 (m, 28H); ¹³C NMR (125 MHz, CDCl₃) δ 167.6, 161.1, 149.2, 138.2, 134.0, 113.6, 101.0, 91.4, 82.0, 75.3, 68.8, 60.0, 51.8, 17.4, 17.3, 17.21, 17.20, 16.98, 16.91, 16.8, 16.7, 13.3, 12.8, 12.7, 12.4. FAB-HRMS M^+ calcd for $C_{25}H_{42}N_2O_9Si_2$ 570.7801, found 570.7804.

3',5'-O-(Tetraisopropyldisiloxane-1,3-diyl)-2'-O-allyl- N^3 -(methoxycarbonylvinyl)uridine (U3). A solution of U2 (3.20 g, 5.60 mmol) and allyl methyl carbonate (1.30 g, 11.20 mmol) in THF (20 mL) was added dropwise via cannula over 15 min, rinsing the flask containing **U2** with THF (3×2 mL), to a suspension of 1,4-bisphosphinobutane (92 mg, 0.20 mmol) and Pd₂(dba)₃ (52 mg, 0.06 mmol) in THF (6 mL). After the mixture was stirred at room temperature for 12 h, the solution was quenched with MeOH (4 mL) and concentrated under vacuum. The residual green viscous oil was purified via chromatography with 20% EtOAc in hexanes as the eluant to produce 2.90 g (85%) of **U3** as a white foam: R_f 0.50 (4:1 hexanes/EtOAc); ¹H NMR (400 MHz, CDCl₃) δ 8.26 (d, J = 15Hz, 1H), 7.93 (d, J = 8.3 Hz, 1H), 7.08 (d, J = 15 Hz, 1H), $5.92 \text{ (ddt, } J = 17.3, \ 10.5, \ 5.2 \text{ Hz, } 1\text{H}), \ 5.74 \text{ (s, } 1\text{H), } 5.73 \text{ (d, }$ J = 8.3 Hz, 1H), 5.39 (dd, J = 17.3, 1.7 Hz, 1H), 5.21 (dd, J = 17.3, 1.7 Hz), 5.21 (dd, J = 1710.5, 1.2 Hz, 1H), 4.39 (m, 2H), 4.25 (ABqd, J = 14 Hz, 1H), 4.19 (ABqd, J = 9.6, 1.6 Hz, 1H), 4.14 (ABqd, J = 9.6, 4.0 Hz,1H), 3.96 (ABqd, J = 14.0, 2.0 Hz, 1H), 3.86 (d, J = 4 Hz, 1H), 3.77 (s, 3H), 0.99-1.10 (m, 28H); ¹³C NMR (125 MHz, CDCl₃) $\delta\ 167.5, 161.1, 149.1, 137.8, 134.13, 134.10, 113.5, 117.6, 100.7,$ 89.6, 81.9, 80.9, 71.2, 67.9, 59.2, 51.7, 17.44, 17.40, 17.25, 17.20, 17.0, 16.9, 16.7, 13.4, 13.0, 12.8, 12.3. FAB-HRMS M⁺ calcd for C₂₈H₄₇N₂O₉Si₂ 611.2821, found 611.2820.

3',5'-O-(Tetraisopropyldisiloxane-1,3-diyl)-2'-O-(2-hydroxyethyl)uridine (U4). A solution of U3 (2.80 g, 4.58 mmol) in CH₂Cl₂ (50 mL) was cooled to −78 °C. Ozone was bubbled through the solution for 20 min until a faint blue color persisted. The ozonide was reduced by the addition of NaBH₄ (0.976 g, 25.8 mmol) dissolved in EtOH (40 mL). After the mixture was stirred for 3.5 h, the solution was quenched with saturated aqueous NH₄Cl (100 mL), and the cloudy white mixture was diluted with EtOAc (200 mL). The organic layer was separated, and the aqueous layer was extracted with EtOAc (2 × 50 mL). The combined organic extracts were washed with saturated aqueous NaCl (75 mL), dried over Na₂-SO₄, and concentrated under vacuum. The residual colorless viscous oil was purified via chromatography with 0-7% MeOH in CH₂Cl₂ as the eluant to yield 1.60 g (62%) of **U4** as a white foam: R_f 0.42 (1:19 MeOH/CH₂Cl₂); ¹H NMR (500 MHz, CDCl₃) δ 9.08 (br s, 1H), 7.89 (d, J = 8.0 Hz, 1H), 5.73 (br s, 1H), 5.69 (dd, J = 8.0, 1.6 Hz, 1H), 4.26 (ABq, J = 13.6 Hz, 1H), 4.19(ABqd, J = 10.0, 4.0 Hz, 1H), 4.15 (ABqd, J = 10.0, 2.0 Hz,1H), 4.03-3.88 (m, 4H), 3.74-3.70 (m, 2H), 2.92 (br s, 1H), 1.10-1.02 (m, 28H); 13 C NMR (125 MHz, CDCl₃) δ 163.2, 150.1, $139.2,\,101.7,\,89.4,\,82.9,\,81.8,\,73.0,\,68.1,\,61.6,\,59.2,\,17.4,\,17.3,$ 17.2, 17.1, 17.0, 16.92, 16.90, 16.7, 13.4, 13.0, 12.8, 12.5. FAB-HRMS $[M\ +\ H]^+$ calcd for $C_{23}H_{43}N_2O_8Si_2\ 531.2558,$ found 531.2557.

3',5'-O-(Tetraisopropyldisiloxane-1,3-diyl)-2'-O-(2-methanesulfonylethyl)uridine (U5). A portion of U4 (2.20 g, 4.14 mmol, combined from two preparations of U4 as described above) was coevaporated with pyridine (10 mL) and dissolved in CH₂Cl₂ (35 mL). The solution was cooled to 0 °C, and Et₃N (0.69 mL, 4.96 mmol) and methanesulfonyl chloride (0.38 mL, 4.96 mmol) were added. After 45 min, the solution was quenched with MeOH (5 mL) and concentrated under vacuum. The residual colorless viscous oil was purified via chromatography with 0-50% EtOAc in CH₂Cl₂ as the eluant to produce 2.10 g (82%) of **U5** as a white foam: R_f 0.36 (1:4 EtOAc/CH₂-Cl₂); ¹H NMR (400 MHz, CDCl₃) δ 9.60 (br s, 1H), 7.89 (d, J =8.0 Hz, 1H), 5.72 (s, 1H), 5.68 (d, J = 8.0 Hz, 1H), 4.45-4.25 (dt, J = 4.2, 2.0 Hz, 2H), 4.24 (ABq, J = 13.6 Hz, 1H), 4.18 (ABqd, J = 9.6, 4.5 Hz, 1H), 4.13-4.09 (m, 3H), 3.95 (ABqd, 3H)J = 13.6, 2.0 Hz, 1H), 3.87 (d, J = 4.0 Hz, 1H), 3.09 (br s, 3H), 1.01 (m, 28H); 13 C NMR (125 MHz, CDCl₃) δ 163.8, 150.2, 139.2, 101.7, 88.5, 82.9, 81.6, 69.4, 68.9, 68.1, 59.2, 37.6, 17.4, 17.3, 17.2, 17.1, 17.0, 16.9, 16.8, 16.7, 13.3, 13.0, 12.8, 12.3. EI-HRMS $[M + H]^+$ calcd for $C_{24}H_{45}N_2O_{10}Si_2S$ 609.2336, found 609.2333.

3',5'-O-(Tetraisopropyldisiloxane-1,3-diyl)-2'-O-(2-azidoethyl)uridine (U6). Na N_3 (0.88 g, 13.5 mmol) and 18crown-6 (5 mg, 0.019 mmol) were added to a solution of U5 $(1.65~\mathrm{g},\,2.71~\mathrm{mmol})$ in DMF $(25~\mathrm{mL})$. The solution was stirred at 60 °C for 6 h. The solvents were removed under vacuum, and the residual colorless viscous oil was partitioned between EtOAc (75 mL) and water (75 mL). The organic layer was separated, and the aqueous layer was extracted with EtOAc $(2 \times 75 \text{ mL})$. The combined organic extracts were washed with saturated aqueous NaCl (75 mL), dried over Na₂SO₄, and concentrated under vacuum to yield 1.57 g (104% of the theoretical amount) of **U6** as a pale yellow foam. Most of this crude sample was used in the next reaction step without further purification: $R_f 0.50 (3:1 \text{ CH}_2\text{Cl}_2/\text{EtOAc}); {}^1\text{H NMR} (400)$ MHz, $CDCl_3$) δ 8.84 (br s, 1H), 7.91 (d, J = 8.0 Hz, 1H), 5.74 (s, 1H), 5.68 (d, J = 8.0 Hz, 1H), 4.25 (ABq, J = 14.0 Hz, 1H), 4.193 (d, J = 4.0 Hz, 1H), 4.190 (m, 1H), 4.03 (ddd, J = 6.4,4.0, 2.8 Hz, 2H), 3.97 (ABqd, J = 13.6, 1.6 Hz, 1H), 3.86 (d,J = 3.2 Hz, 1H, 3.41 (ABqdd, J = 13.2, 6.0, 4.0 Hz, 2H), 1.100.93 (m, 28H); 13 C NMR (100 MHz, CDCl₃) δ 163.3, 149.9, 139.5, 101.6, 88.9, 82.8, 81.6, 70.5, 68.1, 59.3, 50.8, 17.5, 17.4,17.3, 17.2, 17.1, 17.0, 16.8, 16.7, 13.4, 13.0, 12.8, 12.4. EI-HRMS $[M + H]^+$ calcd for $C_{23}H_{42}N_5O_7Si_2$ 556.2623, found 556.2622.

2'-O-(2-(Trifluoroacetamido)ethyl)uridine (U7). Triphenylphosphine (1.38 g, 5.26 mmol) and water (0.24 mL, 13.2

mmol) were added to a solution of crude U6 (1.46 g, assumed 2.63 mmol) dissolved in THF (25 mL). The solution was stirred at 45 °C for 4.5 h. After the solution was cooled to room temperature, ethyl trifluoroacetate (1.56 mL, 13.2 mmol) and Et₃N (1.83 mL, 13.2 mmol) were added, and the mixture continued to be stirred at room temperature for 10 h. The solution was cooled to 0 °C, and 1 M TBAF in THF (5.25 mL, 5.25 mmol) was added. The solution was warmed to room temperature over 5 h, and the solvents were removed under vacuum. The residual colorless viscous oil was purified via chromatography with 5% MeOH in EtOAc as the eluant to produce 0.95 g (94%) of $\mathbf{U7}$ as a white foam: R_f 0.35 (1:19 MeOH/EtOAc); ¹H NMR (400 MHz, acetone- d_6) δ 10.06 (br s, 1H), 8.59 (br s, 1H), 8.07 (d, J = 8.0 Hz, 1H), 5.92 (d, J = 4.0Hz, 1H), 5.57 (d, J=8.0 Hz, 1H), 4.39 (br s, 1H), 4.34 (br t, J = 5.2 Hz, 1H, 4.23 (br s, 1H), 4.13 (dd, J = 4.8, 3.2 Hz, 1H),3.99-3.93 (m, 2H), 3.89 (br s, 1H), 3.84-3.78 (m, 2H), 3.65-3.993.49 (m, 2H); 13 C NMR (125 MHz, acetone- d_6) δ 163.8, 157.9, 151.4, 141.1, 117.2, 102.4, 88.5, 85.7, 83.3, 69.5, 69.3, 61.2, 40.5. 19 F NMR (470 MHz, acetone- d_6) δ –76.9. EI-HRMS [M + H]⁺ calcd for C₁₃H₁₇N₃O₇F₃ 384.1022, found 384.1018.

5' - O - (4, 4' - Dimethoxytrityl) - 2' - O - (2 - (trifluoroacetamido) - (2 - (trifethyl)uridine (U8). 4,4'-Dimethoxytrityl chloride (0.53 g, 1.56 mmol) and DMAP (2 mg, 0.016 mmol) were added to a solution of U7 (0.50 g, 1.3 mmol) in pyridine (10 mL). The solution was stirred at room temperature for 20 h, and the solvents were removed under vacuum. The orange viscous oil was purified by column chromatography. The silica gel was swirled with 10% Et₃N in hexanes (~1 mL of solvent per 1 g of silica gel) for several minutes, and the solvents were removed under vacuum until a freely flowing powder was obtained. The column was packed with this prewashed silica gel using CH₂-Cl₂. After the sample had been loaded, the column was eluted with a gradient of 0-5% MeOH in CH_2Cl_2 to produce 0.67 g (75%) of **U8** as a white foam: $R_f 0.32 (1:19 \text{ MeOH/CH}_2\text{Cl}_2); {}^1\text{H}$ NMR (400 MHz, CDCl₃) δ 8.05 (d, J = 8.0 Hz, 1H), 7.63 (br s, 1H), 7.39-7.37 (m, 2H), 7.32-7.27 (m, 7H), 6.86-6.82 (m, 4H), 5.85 (d, J = 1.2 Hz, 1H), 5.31 (d, J = 8.5 Hz, 1H), 4.42 (dd, 1.85 Hz, 1.85 Hz, 1.85 Hz)J = 8.4, 5.2 Hz, 1H, 4.05 - 4.01 (m, 2H), 3.96 (dd, J = 6.4, 4.0 (m)Hz, 1H), 3.93 (d, J = 6.4 Hz, 1H), 3.79 (s, 6H), 3.64 (br s, 1H), 3.62-3.59 (m, 2H), 3.53 (ABqd, J = 11.2, 2.5 Hz, 1H); 13 C NMR $(125 \text{ MHz}, \text{CDCl}_3) \delta 163.1, 158.73, 158.69, 150.8, 144.2, 139.8,$ 135.2, 135.0, 130.1, 130.0, 128.1, 128.0, 127.2, 113.3, 102.4, 88.2, 87.1, 83.3, 83.1, 69.2, 68.4, 60.8, 55.2, 39.6; ¹⁹F NMR (470 MHz, CDCl₃) δ -76.1. FAB-HRMS [M + H]⁺ calcd for C₃₄H₃₄N₃O₉F₃ 685.2248, found 685.2247.

5'-O-(4,4'-Dimethoxytrityl)-2'-O-(2-(trifluoroacetamido)ethyl)uridine-3'-O-(2-cyanoethyl-N,N-diisopropyl)phos**phoramidite** (U9). A portion of U8 (0.42 g, 0.60 mmol) was coevaporated with pyridine (2 \times 10 mL) and dissolved in CH₂-Cl₂ (10 mL). The solution was cooled to 0 °C, and 2-cyanoethyl N,N,N',N'-tetraisopropylphosphorodiamidite (0.27 mL, 0.82 mmol) and 4,5-dicyanoimidazole²⁷ (DCI, 90 mg, 0.76 mmol) were added. The cloudy mixture was warmed to room temperature, stirred for 10 h, and diluted with $CH_2Cl_2\ (75\ mL)$ and saturated aqueous NaHCO₃ (100 mL). The layers were separated, and the aqueous layer was extracted with CH₂Cl₂ $(3 \times 50 \text{ mL})$. The combined organic extracts were washed with saturated aqueous NaCl (100 mL), dried over Na₂SO₄, and concentrated under vacuum. The residual colorless viscous oil was purified via chromatography; the column was packed with silica gel using 5% Et₃N in hexanes. After the sample had been loaded in CH_2Cl_2 , the column was eluted with 0-3% Et₃N in CH_2Cl_2 to produce 0.44 g (81%) of **U9** as a colorless oil: R_f 0.80 (1:1 EtOAc/CH₂Cl₂). NMR spectroscopy revealed the presence of both diastereomers: ¹H NMR (500 MHz, acetone d_6) δ 8.67 and 8.64 (each br s, total 1H), 7.98 and 7.94 (each d, J = 8.0 Hz, total 1H), 7.52-7.47 (m, 2H), 7.41-7.31 (m, 7H), 7.28-7.24 (m, 1H), 6.93-6.89 (m, 4H), 5.94 and 5.93 (each d, J = 3.0 Hz, total 1H), 5.22 and 5.21 (each d, J = 8.0 Hz, total 1H), 4.67 and 4.59 (each ddd, J = 9.5, 7.0, 5.0 Hz, total 1H), 4.31 and 4.27 (each dd, J = 5.0, 2.5 Hz, total 1H), 4.26 and 4.21 (each td, J=7.5, 3.0 Hz, total 1H), 3.98–3.80 (m, 3H), 3.79 and 3.78 (each s, total 6H), 3.75–3.45 (m, 7H), 2.80 and 2.63 (t with J=6.0 Hz and m, total 2H), 1.20, 1.19, 1.18, 1.17, and 1.08 (each d, J=7.0 Hz, total 12H); 13 C NMR (100 MHz, acetone- d_6) δ 163.5, 159.8, 159.7, 151.3, 145.7, 140.5, 136.3, 136.1, 131.13, 131.10. 129.10, 129.07, 128.7, 127.8, 114.0, 102.4, 89.1, 87.5, 83.07, 83.00, 81.95, 81.90, 70.8, 69.0, 62.2, 59.3, 59.1, 55.5, 43.9, 43.8, 40.3, 25.00, 24.96, 24.90, 24.80, 20.72, 20.70; 19 F NMR (376 MHz, acetone- d_6) δ -76.71 and -76.68; 31 P NMR (202 MHz, acetone- d_6) δ 151.2 and 150.1. FAB-HRMS [M + Na]⁺ calcd for C₄₃H₅₁N₅O₁₀F₃NaP 908.3224, found 908.3223.

5'-O-(4,4'-Dimethoxytrityl)-2'-O-(2-(trifluoroacetamido)ethyl)-4-(1,2,4-triazolyl)uridine-3'-O-(2-cyanoethyl-N,Ndiisopropyl)phosphoramidite (U10). A portion of 1,2,4triazole (0.80 g, 11.6 mmol) was dissolved in acetonitrile (15 mL) and cooled to 0 °C. POCl₃ (200 μL, 2.0 mmol) was added dropwise over 10 min to the cloudy white mixture, and the mixture was stirred at 0 °C for 20 min. A portion of Et₃N (8.0 mL, 57.4 mmol) was added, and the mixture was stirred at 0 °C for 3 h. A solution of U9 (0.200 g, 0.22 mmol) in CH₃CN (0.5 mL) was transferred into the reaction flask via cannula, rinsing the flask containing U9 with CH_3CN (2 \times 0.3 mL). The mixture was stirred at 0 °C for 1.5 h and diluted with 10% Et₃N in EtOAc (25 mL). Saturated aqueous NaHCO₃ (25 mL) was added to the mixture resulting in two clear layers. The organic layer was separated, washed with saturated aqueous NaHCO₃ (25 mL), dried over Na₂SO₄, and concentrated under vacuum. The residual pale yellow foam was purified via chromatography with the column packed as described for U9 (the sample was loaded in 1:4 hexanes/CH₂-Cl₂). The column was eluted with 1:4 hexanes/CH₂Cl₂ containing 3% Et₃N to yield 0.100 g (45%) of **U10** as a white foam: R_f 0.42 (1:5:4 Et₃N/hexanes/CH₂Cl₂). NMR spectroscopy revealed the presence of both diastereomers: ¹H NMR (500 MHz, acetone- d_6) δ 9.241 and 9.240 (each s, total 1H), 8.90 and 8.86 (each d, J = 7.0 Hz, total 1H), 8.23 and 8.22 (each s, total 1H), 7.56 and 7.52 (each d, J = 7.5 Hz, total 2H), 7.45–7.31 (m, 9H), 6.94 (m, 4H), 6.56 and 6.50 (each d, J = 7.5 Hz, total 1H), 5.92 (s, 1H), 4.81 and 4.72 (td, J = 9.5, 4.5 Hz, total 1H), 4.38-4.28 (m, 2H), 4.15 (m, 1H), 4.00 (m, 1H), 3.81 and 3.78 (each s, total 6H), 3.75-3.57 (m, 8H), 2.65 (m, 1H), 1.22 (d, $J = 6.5 \text{ Hz}, 6\text{H}, 1.17 (d, J = 6.5 \text{ Hz}, 6\text{H}); {}^{13}\text{C NMR} (125 \text{ MHz}, 6\text{H}); {}^{13}\text{C NMR} (125 \text{ MHz}, 6\text{Hz}); {}^{13}\text{C NMR} (125 \text{ MHz}, 6\text{Hz});$ acetone- d_6) δ 159.5, 159.2, 154.2, 154.0, 147.6, 144.8, 143.3, 135.9, 135.5, 130.56, 130.46, 128.6, 128.2, 127.3, 118.5, 113.4, 106.8, 94.2, 91.4, 90.9, 87.1, 87.2, 82.2, 81.9, 81.1, 69.2, 69.1, 68.7, 60.5, 60.3, 58.6, 58.4, 54.9, 46.3, 43.3, 43.2, 43.1, 39.7, 39.6, 39.5, 26.1, 24.3, 20.2; $^{19}{\rm F}$ NMR (470 MHz, acetone- d_6) δ -76.75 and -76.72; ³¹P NMR (202 MHz, acetone- d_6) δ 151.7 and 149.5. ESI-HRMS $[M + H]^+$ calcd for $C_{45}H_{53}N_8O_9F_3P$ 937.3625, found 937.3594.

3',5'-O-(Tetraisopropyldisiloxane-1,3-diyl)-2'-O-(2-(tri**phenylmethylthio)ethyl)uridine** (U11). Diisopropyl azodicarboxylate (1.1 mL, 5.6 mmol) was added to a solution of triphenylphosphine (1.48 g, 5.6 mmol) in THF (30 mL) cooled to 0 °C. After 0.5 h, a solution of U4 (1.50 g, 2.82 mmol) and trityl mercaptan (1.55 g, 5.6 mmol) in THF (20 mL) was added via cannula, and the flask containing the latter reagents was rinsed with THF (2 \times 5 mL). After 40 min, the solution was quenched with MeOH (2 mL) and concentrated under vacuum. The resulting colorless viscous oil was purified via chromatography with 10-30% EtOAc in hexanes as the eluant to produce 1.90 g (85%) of **U11** as a white foam: R_f 0.35 (3:1) hexanes/EtOAc); 1 H NMR (500 MHz, CDCl₃) δ 9.43 (s, 1H), 7.88 (d, J = 8.0 Hz, 1H), 7.42 - 7.18 (m, 15H), 5.67 (dd, J = 8.0 Hz, 1H)6.0, 2.0 Hz, 1H), 5.66 (s, 1H), 4.22 (ABq, J = 13.5 Hz, 1H), 4.12 (s, 1H), 3.95 (dd, J = 13.5, 2.0 Hz, 1H), 3.76-3.68 (m, 4H), 2.47-2.44 (m, 2H), 1.10-0.97 (m, 28H); ¹³C NMR (125 MHz, $CDCl_3$) δ 164.1, 150.0, 145.1, 139.9, 129.9, 128.1, 126.8, 101.6, 89.2, 82.6, 81.8, 70.0, 68.4, 66.7, 59.6, 32.2, 17.7, 17.6,

17.5, 17.4, 17.3, 17.2. 17.1. 17.0, 13.6, 13.3, 13.1, 12.7. ESI-HRMS $[M+Na]^+$ calcd for $C_{42}H_{56}N_2O_7SSi_2Na$ 811.3245, found 811.3296.

2'-O-(2-(Triphenylmethylthio)ethyl)uridine (U12). TBAF (1 M) in THF (5.6 mL, 5.6 mmol) was added dropwise over 5 min to a solution of U11 (1.90 g, 2.41 mmol) in THF (30 mL) at 0 °C. The solution was stirred at room temperature for 6 h, diluted with EtOAc (20 mL), washed with saturated aqueous $NaHCO_3$ (2 × 25 mL), dried over Na_2SO_4 , and concentrated under vacuum. The resulting colorless viscous oil was purified via chromatography with 0−5% MeOH in CH₂Cl₂ as the eluant to produce 0.96 g (72%) of **U12** as a white foam: R_f 0.33 (1:19 MeOH/CH₂Cl₂); ¹H NMR (400 MHz, CDCl₃) δ 9.81 (br s, 1 H), $7.75 \, (d, J = 8.0 \, Hz, 1H), 7.41 - 7.19 \, (m, 15H), 5.69 \, (d, J = 8.0 \, Hz)$ Hz, 1H), 5.65 (d, J = 4.0 Hz, 1H), 4.20 (t, J = 5.2 Hz, 1H), 4.01 (br dt, J = 5.6, 2.4 Hz, 1H), 3.94 (dd, J = 5.2, 3.5 Hz, 1H), 3.93 (ABqd, J = 12.2, 1.5 Hz, 1H), 3.77 (ABqd, J = 11.2, 2.0 Hz, 1H, 3.60 (dt, J = 10.0 Hz, 5.6 Hz, 1H), 3.38 (br s, 1H), $3.23 \, (ddd, J = 10.5, 8.0, 5.0 \, Hz, 1H), 3.12 \, (br \, s, 1H), 2.54 \, (ddd, J \, s, 1H), 2.54$ $J = 13.0, 7.4, 5.4 \text{ Hz}, 1\text{H}), 2.40 (dt, <math>J = 13.0, 6.0 \text{ Hz}, 1\text{H}); {}^{13}\text{C}$ NMR (100 MHz, CDCl₃) δ 163.6, 150.4, 144.7, 142.2, 129.8, 128.3, 127.1, 102.6, 90.7, 85.3, 81.2, 69.6, 68.9, 61.6, 32.3. ESI-HRMS $[M + Na]^+$ calcd for $C_{30}H_{30}N_2O_6SNa$ 569.1722, found 569.1714.

5'-O-(4,4'-Dimethoxytrityl)-2'-O-(2-(triphenylmethylthio)ethyl)uridine (U13). A portion of U12 (0.90 g, 1.65 mmol) was coevaporated from pyridine (3 mL) and dissolved in pyridine (3 mL). The solution was cooled to 0 °C, and 4,4'dimethoxytrityl chloride (1.00 g, 2.95 mmol) was added. After the mixture was stirred at room temperature for 12 h, the solvents were removed under vacuum. The residual orange foam was purified via chromatography with the column packed as described for U9. The column was eluted with a gradient of 0-5% MeOH in CH_2Cl_2 to produce 0.95 g (68%) of **U13** as a pale yellow foam: $R_f 0.45$ (1:19 MeOH/CH₂Cl₂); ¹H NMR (500 MHz, acetone- d_6) δ 10.07 (br s, 1H), 7.733 (s, 1H), 7.85 (d, J =8.5 Hz, 1H), 7.48-7.21 (m, 24H), 6.92-6.90 (m, 4H), 5.89 (d, J = 3.0 Hz, 1H), 5.26 (d, J = 8.0 Hz, 1H), 4.41 (q, J = 5.5 Hz,1H), 4.06 (dt, J = 6.5, 3.0 Hz, 1H), 3.95 (dd, J = 4.5, 3.0 Hz, 1H), 3.71 (br q, J = 7.5 Hz, 1H), 3.65 (ABqd, J = 10.5, 7.0 Hz, 1H), 3.54 (ABqt, J = 10.5, 6.5 Hz, 1H), 3.79 (s, 6H), 3.47 (ABqd, J = 11.0, 3.0 Hz, 1H), 3.34 (ABqd, J = 11.0, 3.0 Hz, 1H), 2.86(br s, 1H); $^{13}\mathrm{C}$ NMR (125 MHz, acetone- d_{6}) δ 163.0, 159.1, 150.6, 145.2, 135.9, 135.7, 130.4, 129.8, 128.4, 128.2, 128.1, 127.1, 126.9, 113.4, 101.8, 87.8, 86.8, 83.3, 82.4, 69.1, 67.1, 62.6, 54.9, 32.0, 18.5. ESI-HRMS $[M\,+\,Na]^+$ calcd for $C_{51}H_{48}N_2O_{8^-}$ SNa 871.3029, found 871.3060.

5'-O-(4,4'-Dimethoxytrityl)-2'-O-(2-(triphenylmethylthio)ethyl)uridine-3'-O-(2-cyanoethyl-N,N-diisopropyl)phosphoramidite (U14). A portion of U13 (0.30 g, 0.35 mmol) was coevaporated with pyridine (2 × 5 mL) and dissolved in CH₂Cl₂ (4 mL). The solution was cooled to 0 °C, and 2-cyanoethyl N,N,N',N'-tetraisopropylphosphorodiamidite (0.15 mL, 0.45 mmol) and 4,5-dicyanoimidazole (50 mg, 0.42 mmol) were added. The cloudy mixture was warmed to room temperature and stirred for 6 h. The mixture was diluted with 5% Et₃N in EtOAc (25 mL) and washed with saturated aqueous NaHCO₃ $(2 \times 25 \text{ mL})$. The organic layer was dried over Na₂SO₄ and concentrated under vacuum. The residual pale yellow foam was purified via chromatography with the column packed as described for U9 (the sample was loaded in 1:15:4 Et₃N/ hexanes/CH₂Cl₂). The column was eluted with a gradient from $1{:}15{:}4$ to $1{:}7{:}12$ $Et_3N/hexanes/CH_2Cl_2$ to produce 0.31 g (84%)of **U14** as a white foam: R_f 0.51 (1:7:12 Et₃N/hexanes/CH₂-Cl₂). NMR spectroscopy revealed the presence of both diastereomers: ¹H NMR (500 MHz, acetone- d_6) δ 10.10 (br s, 1H), 7.93 and 7.86 (each d, J = 8.0 Hz, total 1H), 7.53-7.22 (m, 24H), 6.87 (m, 4H), 5.93 and 5.91 (each d, J = 3.0 Hz, total 1H), 5.23 and 5.21 (each d, J = 2.5 Hz, total 1H), 4.57 and 4.49 (each m, total 1H), 4.21 and 4.15 (each m, total 1H), 4.08 (m, 1H), 3.88 (m, 1H), 3.79 and 3.78 (each s, total 6H), 3.68 (m, 1H), 3.63 (m, 4H), 3.55-3.43 (m, 2H), 2.71 (m, 1H), 2.62 (m, 1H), 2.45 (m, 2H), 1.18 (m, 12H); ^{13}C NMR (125 MHz, acetone- d_6) δ 162.5, 159.0, 150.5, 145.2, 140.2, 130.5, 129.8, 128.4, 128.3, 127.1, 126.9, 113.3, 101.7, 88.0, 87.0, 82.2, 82.3, $82.1, 70.3, 69.3, 69.1, 66.2, 62.2, 43.2, 32.1, 24.2, 20.1; {}^{31}\mathrm{P}\,\mathrm{NMR}$ (202 MHz, acetone- d_6) δ 150.8 and 150.4. ESI-HRMS [M + H]⁺ calcd for $C_{60}H_{66}N_4O_9SP$ 1049.4288, found 1049.4281.

5'-O-(4,4'-Dimethoxytrityl)-2'-O-(2-(triphenylmethylthio)ethyl)-4-(1,2,4-triazolyl)uridine-3'-O-(2-cyanoethyl-N,N-diisopropyl)phosphoramidite (U15). A portion of 1,2,4-triazole (0.65 g, $9.\overline{4}$ mmol) was dissolved in acetonitrile (13 mL) and cooled to 0 °C. POCl₃ (165 μ L, 1.8 mmol) was added dropwise over 10 min to the cloudy white mixture, and the mixture was stirred at 0 °C for 20 min. A portion of Et₃N (5.1 mL, 36.6 mmol) was added, and the mixture was stirred at 0 °C for 3 h. A solution of **U14** (0.197 g, 0.188 mmol) in CH₃CN (1.5 mL) was transferred into the reaction flask via cannula; the flask containing U14 was rinsed with CH3CN $(2 \times 0.2 \text{ mL})$. The mixture was stirred at 0 °C for 1.5 h and diluted with 10% Et₃N in EtOAc (25 mL). Saturated aqueous NaHCO₃ (25 mL) was added to the mixture resulting in two clear layers. The organic layer was separated, washed with saturated aqueous NaHCO3 (25 mL), dried over Na2SO4, and concentrated under vacuum. The residual pale yellow foam was purified via chromatography with the column packed as described for U9 (the sample was loaded in 3:2 hexanes/CH2-Cl₂). The column was eluted with 1:11:8 to 1:8:11 Et₃N/ hexanes/CH₂Cl₂ to yield 0.112 g (54%) of U15 as a white foam: R_f 0.55 (1:11:8 Et₃N/hexanes/CH₂Cl₂). NMR spectroscopy revealed the presence of both diastereomers: ¹H NMR (500 MHz, acetone- d_6) δ 9.24 and 9.23 (each s, total 1H), 8.89 and 8.83 (each d, J = 7.0 Hz, total 1H), 8.22 and 8.21 (each s, total 1H), 7.57-7.23 (m, 24H), 6.93 (m, 4H), 6.53 and 6.46 (each d, J = 7.0 Hz, total 1H), 5.91 and 5.90 (each s, total 1H), 4.77 and 4.64 (each td, J = 9.0, 5.0 Hz, total 1H), 4.30 (m, 1H), 4.15 (ABqd, J = 10.5, 4.5 Hz, 1H), 3.96-3.82 (m, 2H), $3.81 \ and \ 3.80 \ (each \ s, \ total \ 6H), \ 3.78-3.56 \ (m, \ 6H), \ 2.85 \ and$ 2.82 (each s, total 1H), 2.68 (m, 1H), 2.62 (m, 1H), 2.53 (m, 1H), 1.19 (d, J = 7.5 Hz, 3H), 1.14 (d, J = 7.0 Hz, 3H), 1.10 (d, J = 7.0 Hz, 3H) $J=7.0~{\rm Hz},\,3{\rm H}).\,\,1.05~({\rm d},\,J=6.5~{\rm Hz},\,3{\rm H});\,^{13}{\rm C}$ NMR (125 MHz, acetone- d_6) δ 159.4, 159.1, 154.2, 147.7, 145.3, 144.8, 143.2, 135.5, 135.4, 130.6, 130.5, 130.4, 130.3, 129.8, 128.6, 128.2, 127.3, 126.9, 113.3, 94.0, 90.7, 87.1, 82.2, 69.1, 68.9, 60.5, 60.3, $58.6, 58.4, 54.9, 43.3, 43.2, 32.1, 24.5, 20.0, 18.6; {}^{31}P\ NMR\ (202$ MHz, acetone- d_6) δ 151.2 and 149.6. ESI-HRMS [M + Na]⁺ calcd for C₆₂H₆₆N₇O₈PSNa 1122.4329, found 1122.4363.

2'-O-(Methoxycarbonylmethyl)adenosine (A2). A 60% dispersion of sodium hydride in mineral oil (1.26 g, 32 mmol) was added to a suspension of adenosine (6.15 g, 23.0 mmol) in DMF (150 mL) at $\hat{0}$ °C. The suspension was stirred at 0 °C for 1 h, and methyl bromoacetate (3.3 mL, 34.5 mmol) was added. The mixture was warmed to room temperature, stirred for 8 h, and quenched with acetic acid (3 mL) and methanol (10 mL). The solvents were removed under vacuum, and the resulting paste was purified via chromatography (dry-load) with 1-10% MeOH in CH_2Cl_2 as the eluant to produce 3.65 g (47%) of **A2** as a white foam: R_f 0.35 (1:9 MeOH/CH₂Cl₂); 1 H NMR (400 MHz, CDCl₃) δ 8.32 (s, 1H), 7.87 (s, 1H), 6.58 (br s, 1H), 5.96 (br s, 2H), 5.94 (d, J = 7.5 Hz, 1H), 4.73 (dd, J = 7.6, 4.3 Hz, 1H), 4.43 (d, J = 5.5 Hz, 1H), 4.40-4.31 (m, 4H), 3.97-3.91(m, 2H), 3.74 (s, 3H), 3.72 (br s, 1H); ¹³C NMR (125 MHz, $CDCl_3$) δ 172.2, 156.1, 152.3, 148.3, 141.0, 121.1, 89.0, 87.4, 84.6, 70.8, 69.1, 63.2, 52.5. ESI-HRMS [M + H]⁺ calcd for $C_{13}H_{18}N_5O_6$ 340.1257, found 340.1266. The HMBC spectrum was also recorded (see text).

3',5'-Bis-O-(tert-butyldimethylsilyl)-2'-O-(methoxycarbonylmethyl)adenosine (A3). tert-Butyldimethylsilyl chloride (5.08 g, 33.7 mmol) and imidazole (2.27 g, 16.2 mmol) were added to a stirred solution of A2 (3.65 g, 10.8 mmol) in DMF (50 mL). After 16 h, the solvents were removed under vacuum, and the resulting paste was partitioned between water (100 mL) and CH₂Cl₂ (100 mL). The aqueous layer was extracted with CH_2Cl_2 (3 × 70 mL). The combined organic extracts were washed with saturated aqueous NaCl (100 mL), dried over MgSO₄, and concentrated under vacuum. The resulting oil was purified via chromatography with 0–3% MeOH in CH₂Cl₂ as the eluant to produce 5.67 g (93%) of **A3** as a white solid: R_f 0.54 (1:9 MeOH/CH₂Cl₂); $^1\mathrm{H}$ NMR (400 MHz, CDCl₃) δ 8.34 (s, 1H), 8.20 (s, 1H), 6.22 (d, 1H, J=4.1 Hz), 5.78 (br s, 2H), 4.58 (dd, J=4.6, 3.5 Hz, 1H), 4.57 (dd, J=4.7, 3.3 Hz, 1H), 4.28 (ABq, J=16.7 Hz, 1H), 4.20 (ABq, J=16.8 Hz, 1H), 4.14 (q, J=3.4 Hz, 1H), 3.96 (ABqd, J=11.4, 3.6 Hz, 1H), 3.77 (ABqd, J=11.5, 3.1 Hz, 1H), 3.64 (s, 3H), 0.93 (s, 9H), 0.91 (s, 9H), 0.14 (s, 3H), 0.13 (s, 3H), 0.09 (s, 3H), 0.08 (s, 3H); $^{13}\mathrm{C}$ NMR (125 MHz, CDCl₃) δ 170.1, 155.3, 152.9, 149.7, 139.7, 120.1, 86.7, 85.1, 82.1, 70.2, 67.6, 61.9, 51.2, 25.9, 25.7, 18.4, 18.1, -4.7, -4.9, -5.4, -5.5. FAB-HRMS [M+H]+ calcd for C₂₅H₄₆N₅O₆Si₂ 568.2984, found 568.2986.

3',5'-Bis-O-(tert-butyldimethylsilyl)-2'-O-(2-hydroxyethyl)adenosine (A4). Lithium aluminum hydride (0.99 g, 25 mmol) was added to a solution of A3 (5.67 g, 9.98 mmol) in ether (45 mL) cooled to 0 °C. After 2 h, the solution was quenched at 0 °C by the slow addition of saturated aqueous NH₄Cl (100 mL). After the solution was warmed to room temperature, it was diluted with ether (100 mL). The aqueous layer was extracted with ether (6 \times 100 mL). The combined organic extracts were washed with saturated aqueous NaCl (200 mL), dried over MgSO₄, and concentrated under vacuum to produce 5.07 g (94%) of **A4** as a colorless oil: R_f 0.61 (1:9 MeOH/CH₂Cl₂); ¹H NMR (400 MHz, CDCl₃) δ 8.29 (s, 1H), 8.19 (s, 1H), 6.18 (d, J = 3.4 Hz, 1H), 6.04 (br s, 2H), 4.49 (dd, J =5.4, 5.0 Hz, 1H), 4.32 (dd, J=4.6, 3.6 Hz, 1H), 4.12 (dd, J=4.6, 3.8 Hz, 1H), 4.12 (dd, 5.80, 2.8 Hz, 1H), 3.99 (ABqd, J = 11.6, 3.2 Hz, 1H), 3.80- $3.70\ (m,\,5H),\,3.65\ (br\ s,\,1H),\,0.92\ (s,\,9H),\,0.91\ (s,\,9H),\,0.10\ (s,\,9H),\,$ 9H), 0.06 (s, 3H). 13 C NMR (100 MHz, CDCl₃) δ 155.5, 152.9, 149.1, 138.9, 119.9, 87.5, 84.6, 82.9, 72.6, 69.6, 61.7, 61.5, 26.0, 25.7, 18.4, 18.0, -4.6, -4.8, -5.4, -5.5. EI-HRMS M⁺ calcd for C₂₄H₄₅N₅O₅Si₂ 539.2959, found 539.2953.

3',5'-Bis-O-(tert-butyldimethylsilyl)-2'-O-(2-hydroxyethyl)-N⁶-isobutyryladenosine (A5). Trimethylsilyl chloride (3.0 mL, 23.6 mmol) was added to a solution of A4 (2.41 g, 4.46 mmol) in pyridine (20 mL) resulting in a cloudy mixture. After 1 h, isobutyric anhydride (3.0 mL, 18.1 mmol) was added. After an additional 5 h, the mixture was cooled to 0 °C, and water (4 mL) was added. After an additional 30 min, concentrated NH₄OH (4 mL) was added, and the sample was concentrated under vacuum. The resulting paste was partitioned between CH₂Cl₂ (100 mL) and water (100 mL). The aqueous layer was extracted with $CH_{2}Cl_{2}$ (3 \times 100 mL). The combined organic extracts were washed with saturated aqueous NaHCO₃ (50 mL) and saturated aqueous NaCl (50 mL), dried over MgSO₄, and concentrated under vacuum. The residual colorless viscous oil was purified via chromatography with 40-80% EtOAc in $\mathrm{CH_2Cl_2}$ as the eluant to yield 2.44 g (90%) of **A5** as a white foam: R_f 0.39 (4:1 EtOAc/CH₂Cl₂); ¹H NMR (500 MHz, CDCl₃) δ 8.70 (s, 1H), 8.43 (br s, 1H), 8.37 (s, 1H), 6.22 (d, J = 3.5 Hz, 1H), 4.51 (dd, J = 5.6, 5 Hz, 1H), 4.34 (dd, J = 4.3, 3.3 Hz, 1H), 4.15 (dt, J = 4.5, 2.5 Hz, 1H),4.02 (ABqd, J = 11.6, 2.8 Hz, 1H), 3.81-3.74 (m, 5H), 3.24(septet, J = 6.7 Hz, 1H), 2.55 (br s, 1H), 1.32 (d, J = 6.7 Hz, 3H), 1.30 (d, J = 6.7 Hz, 3H), 0.93 (s, 9H), 0.92 (s, 9H), 0.12 (s, 6H), 0.11 (s, 6H); 13 C NMR (100 MHz, CDCl₃) δ 176.3, 152.7, $150.4,\,149.7,\,141.0,\,121.8,\,87.6,\,84.7,\,83.0,\,72.4,\,69.3,\,61.7,\,61.3,\\$ 36.0, 26.0, 25.6, 19.2, 18.8, 18.4, 18.0, -4.7, -4.9, -5.4, -5.6. EI-HRMS M^+ calcd for $C_{28}H_{51}N_5O_6Si_2$ 609.3377, found.

3′,5′-Bis-O-(tert-butyldimethylsilyl)-2′-O-(2-methane-sulfonylethyl)-N6-isobutyryladenosine (A6). Et₃N (0.7 mL, 5 mmol) was added to a solution of A5 (0.61 g, 1.0 mmol) in CH₂Cl₂ (10 mL) cooled to 0 °C and followed by the addition of methanesulfonyl chloride (0.23 mL, 3.0 mmol). After 1 h, the solution was diluted with CH₂Cl₂ (50 mL), and washed with saturated aqueous NaHCO₃ (50 mL). The aqueous layer was extracted with CH₂Cl₂ (3 \times 25 mL), and the combined organic extracts were washed with saturated aqueous NaCl (50 mL),

dried over MgSO₄, and concentrated under vacuum. The residual yellow viscous oil was purified via chromatography with 1:1 EtOAc/CH₂Cl₂ as the eluant to produce 0.57 g (82%) of **A6** as a colorless oil: R_f 0.52 (1:1 EtOAc/CH₂Cl₂); $^1\mathrm{H}$ NMR (500 MHz, CDCl₃) δ 8.84 (br s, 1H), 8.67 (s, 1H), 8.36 (s, 1H), 6.16 (d, J=3.5 Hz, 1H), 4.52 (dd, J=5.3, 4.8 Hz, 1H), 4.34 (dd, J=4.5, 3.8 Hz, 1H), 4.33 (t, J=4.0 Hz, 2H), 4.09 (ddd, J=6.2, 3.0, 2.4 Hz, 1H), 3.98 (ABqd, J=11.6, 3.3 Hz, 1H), 3.89 (m, 2H), 3.76 (ABqd, J=11.6, 2.6 Hz, 1H), 3.16 (m, 1H), 2.98 (s, 3H), 1.27 (d, J=6.8 Hz, 3H), 1.27 (d, J=6.8 Hz, 3H), 0.90 (s, 9H), 0.89 (s, 9H), 0.09 (s, 3H), 0.08 (s, 3H), 0.07 (s, 3H), 0.07 (s, 3H), 1.30 (s, 3H), 1.30

3',5'-Bis-O-(tert-butyldimethylsilyl)-2'-O-(2-azidoethyl)- N^6 -isobutyryladenosine (A7). NaN₃ (0.26 g, 4.0 mmol) was added to a solution of A6 (0.55 g, 0.80 mmol) in DMF (8 mL), and the mixture was heated at 50 $^{\circ}\mathrm{C}$ for 3 h. The solvents were removed under vacuum, and the resulting pale orange residue was partitioned between CH₂Cl₂ (100 mL) and water (100 mL). The aqueous layer was extracted with CH_2Cl_2 (2 × 50 mL). The combined organic extracts were washed with saturated aqueous NaCl (50 mL), dried over MgSO₄, and concentrated under vacuum to produce 0.49 g (97%) of A7 as a colorless oil: R_f 0.78 (1:1 EtOAc/CH₂Cl₂); ¹H NMR (400 MHz, CDCl₃) δ 8.71 (s, 1H), 8.45 (br s, 1H) 8.38 (s, 1H), 6.20 (d, J=4.0 Hz, 1H, 4.49 (dd, J = 5.3, 4.9 Hz, 1H), 4.37 (dd, J = 4.4,3.8 Hz, 1H), 4.16, (ddd, J = 5.9, 3.3, 2.9 Hz, 1H), 4.03 (ABqd, J = 5.9, 3.3, 2.9 Hz, 1H), 4.03 (ABqd, J = 5.9, 3.3, 2.9 Hz, 1H), 4.03 (ABqd, J = 5.9, 3.3, 2.9 Hz, 1H), 4.03 (ABqd, J = 5.9, 3.3, 2.9 Hz, 1H), 4.03 (ABqd, J = 5.9, 3.3, 2.9 Hz, 1H), 4.03 (ABqd, J = 5.9, 3.3, 2.9 Hz, 1H), 4.03 (ABqd, J = 5.9, 3.3, 2.9 Hz, 1H), 4.03 (ABqd, J = 5.9, 3.3, 2.9 Hz, 1H), 4.03 (ABqd, J = 5.9, 3.9), 4.03 (ABqd, J = 5.9, 3.9, 3.9 Hz, 1H), 4.03 (ABqd, J = 5.9, 3.9, 3.9 Hz, 1 Hz, 1 Hz), 4.03 (ABqd, J = 5.9, 3.9, 3.9 Hz, 1 Hz), 4.03 (ABqd, J = 5.9, 3.9, 3.9 Hz, 1 Hz), 4.03 (ABqd, J = 5.9, 3.9, 3.9 Hz, 1 Hz), 4.03 (ABqd, J = 5.9, 3.9, 3.9 Hz, 1 Hz), 4.03 (ABqd, J = 5.9, 3.9, 3.9 Hz), 4.03 (ABqd, J = 5J = 11.5, 3.5 Hz, 1H, 3.84 - 3.78 (m, 3H), 3.45 (ABqt, J = 13.6,5.1 Hz, 1H, 3.37 (ABQt, J = 13.4, 4.8 Hz, 1H), 3.21 (br septet,J = 6.7 Hz, 1H, 1.31 (d, J = 6.8 Hz, 6H), 0.94 (s, 9H), 0.93 (s, 9H)9H), 0.11 (s, 6H,), 0.12 (s, 3H), 0.11 (s, 3H); ¹³C NMR (125 MHz, CDCl₃) δ 176.2, 152.4, 150.7, 149.3, 141.6, 122.6, 87.1, 84.6, 82.2, 69.6, 69.6, 61.4, 50.6, 35.8, 25.8, 25.5, 19.0, 18.2, 17.9, -4.8, -5.1, -5.6. FAB-HRMS [M + H]⁺ calcd $C_{28}H_{51}N_8O_{5-}$ Si₂ 635.3520, found 635.3523.

3',5'-Bis-O-(tert-butyldimethylsilyl)-2'-O-(2-(trifluoroacetamido)ethyl)-N⁶-isobutyryladenosine (A8). Pd/C (10%, 0.20 g, 0.19 mmol) was added to a solution of A7 (0.48 g, 0.76 mmol) in MeOH (9 mL). The flask was evacuated and placed under an atmosphere of hydrogen. The consumption of A7 was monitored by TLC. After 1.5 h, ethyl trifluoroacetate (1.0 mL, $8.4\ mmol)$ and Et₃N (0.7 mL, 5.0 mmol) were added. After 4 h, silica gel (3 g) was added for dry loading, and the solvents were removed under vacuum. The sample was purified via chromatography with 1:4 EtOAc/CH2Cl2 as the eluant to produce 0.44 g (81%) of $\bf A8$ as a pale yellow oily solid: R_f 0.75 (1:1 EtOAc/CH₂Cl₂); ¹H NMR (500 MHz, CDCl₃) δ 9.17 (br s, 1H), 8.62 (s, 1H), 8.35 (s, 1H), 7.38 (br t, J = 4.8 Hz, 1H), 6.12 $(\mathrm{d}, J = 3.6~\mathrm{Hz}, \, 1\mathrm{H}), \, 4.48~(\mathrm{dd}, J = 5.3, \, 4.9~\mathrm{Hz}, \, 1\mathrm{H}), \, 4.28~(\mathrm{dd}, J = 5.3, \, 4.9~\mathrm{Hz}, \, 1\mathrm{H}), \, 4.28~(\mathrm{dd}, J = 5.3, \, 4.9~\mathrm{Hz}, \, 1\mathrm{H}), \, 4.28~(\mathrm{dd}, J = 5.3, \, 4.9~\mathrm{Hz}, \, 1\mathrm{H}), \, 4.28~(\mathrm{dd}, J = 5.3, \, 4.9~\mathrm{Hz}, \, 1\mathrm{H}), \, 4.28~(\mathrm{dd}, J = 5.3, \, 4.9~\mathrm{Hz}, \, 1\mathrm{Hz}), \, 4.28~(\mathrm{dd}, J = 5.3, \, 4.9~\mathrm{Hz}, \, 1\mathrm{Hz}), \, 4.28~(\mathrm{dd}, J = 5.3, \, 4.9~\mathrm{Hz}, \, 1\mathrm{Hz}), \, 4.28~(\mathrm{dd}, J = 5.3, \, 4.9~\mathrm{Hz}, \, 1\mathrm{Hz}), \, 4.28~(\mathrm{dd}, J = 5.3, \, 4.9~\mathrm{Hz}, \, 1\mathrm{Hz}), \, 4.28~(\mathrm{dd}, J = 5.3, \, 4.9~\mathrm{Hz}, \, 1\mathrm{Hz}), \, 4.28~(\mathrm{dd}, J = 5.3, \, 4.9~\mathrm{Hz}, \, 1\mathrm{Hz}), \, 4.28~(\mathrm{dd}, J = 5.3, \, 4.9~\mathrm{Hz}, \, 1\mathrm{Hz}), \, 4.28~(\mathrm{dd}, J = 5.3, \, 4.9~\mathrm{Hz}, \, 1\mathrm{Hz}), \, 4.28~(\mathrm{dd}, J = 5.3, \, 4.9~\mathrm{Hz}, \, 1\mathrm{Hz}), \, 4.28~(\mathrm{dd}, J = 5.3, \, 4.9~\mathrm{Hz}, \, 1\mathrm{Hz}), \, 4.28~(\mathrm{dd}, J = 5.3, \, 4.9~\mathrm{Hz}, \, 1\mathrm{Hz}), \, 4.28~(\mathrm{dd}, J = 5.3, \, 4.9~\mathrm{Hz}, \, 1\mathrm{Hz}), \, 4.28~(\mathrm{dd}, J = 5.3, \, 4.9~\mathrm{Hz}, \, 1\mathrm{Hz}), \, 4.28~(\mathrm{dd}, J = 5.3, \, 4.9~\mathrm{Hz}, \, 1\mathrm{Hz}), \, 4.28~(\mathrm{dd}, J = 5.3, \, 4.9~\mathrm{Hz}, \, 1\mathrm{Hz}), \, 4.28~(\mathrm{dd}, J = 5.3, \, 4.9~\mathrm{Hz}, \, 1\mathrm{Hz}), \, 4.28~(\mathrm{dd}, J = 5.3, \, 4.9~\mathrm{Hz}, \, 1\mathrm{Hz}), \, 4.28~(\mathrm{dd}, J = 5.3, \, 4.9~\mathrm{Hz}, \, 1\mathrm{Hz}), \, 4.28~(\mathrm{dd}, J = 5.3, \, 4.9~\mathrm{Hz}, \, 1\mathrm{Hz}), \, 4.28~(\mathrm{dd}, J = 5.3, \, 4.9~\mathrm{Hz}, \, 1\mathrm{Hz}), \, 4.28~(\mathrm{dd}, J = 5.3, \, 4.9~\mathrm{Hz}), \, 4.28~(\mathrm$ J = 4.4, 3.9 Hz, 1H), 4.08 (ddd, J = 5.3, 2.7, 2.6 Hz, 1H), 3.94 (ABqd, J = 11.6, 3.1 Hz, 1H), 3.80 - 3.75 (m, 1H), 3.73 (ABqd, 1H), 3.80 - 3.75 (m, 1H), 3.73 (ABqd, 1H),J = 11.7, 2.5 Hz, 1H, 3.72 - 3.68 (m, 1H), 3.60 - 3.50 (m, 1H),3.50-3.40 (m, 1H), 3.20 (m, 1H), 1.23 (d, J = 6.9 Hz, 3H), 1.22(d, J = 6.8 Hz, 3H), 0.86 (s, 9H), 0.84 (s, 9H), 0.06 (s, 3H),0.04 (s, 3H), 0.04 (s, 3H), 0.03 (s, 3H); ¹³C NMR (125 MHz, $CDCl_3$) δ 176.4, 157.2, 152.3, 150.6, 149.4, 141.3, 122.5, 115.6, $87.3,\ 84.9,\ 82.4,\ 69.9,\ 68.1,\ 61.4,\ 39.6,\ 35.8,\ 25.8,\ 25.4,\ 19.0,$ 18.2, 17.8, -4.9, -5.1, -5.6, -5.7; ¹⁹F NMR (470 MHz, CDCl₃) δ –76.2. FAB-HRMS [M + H]+ calcd for $C_{30}H_{52}N_6O_6F_3Si_2$ 705.3439, found 704.3437.

2'-O-(2-(Trifluoroacetamido)ethyl)-N6-isobutyryladenosine (A9). TBAF (1 M) in THF (1.4 mL, 1.36 mmol) was added dropwise over 5 min to a solution of A8 (0.440 g, 0.62 mmol) in THF (7 mL) cooled to 0 °C. The solution was warmed to room temperature. After 3 h, silica gel (2 g) was added for dry loading, and the solvents were removed under vacuum. The sample was purified via chromatography with 0-6% MeOH in CH₂Cl₂ as the eluant to produce 0.253 g (86%) of A9 as a clear colorless oil: R_f 0.17 (1:1 EtOAc/CH₂Cl₂); ¹H NMR (500 MHz, CDCl₃) δ 8.69 (s, 1H), 8.65 (br s, 1H), 8.12 (s, 1H),

6.00 (d, J = 6.9 Hz, 1H), 5.87 (br s, 1H), 4.80 (dd, J = 6.9, 4.7)Hz, 1H), 4.64 (br d, J = 3.8 Hz, 1H), 4.36 (d, J = 1.6 Hz, 1H), 3.99 (d, J = 12.9 Hz, 1H), 3.80 (d, J = 12.5 Hz, 1H), 3.70-3.63 (m, 2H), 3.60-3.54 (m, 1H), 3.47-3.40 (m, 1H), 3.26 (septet, J = 6.1 Hz, 1H), 1.32 (d, J = 6.8 Hz, 3H), 1.31 (d, J =6.8 Hz, 3H); ¹³C NMR (125 MHz, CDCl₃) δ 177.3, 157.7, 151.9, 150.3, 149.8, 143.1, 123.3, 115.8, 89.1, 87.7, 81.7, 70.1, 68.5, 62.4, 39.9, 35.9, 19.0; ¹⁹F NMR (470 MHz, CDCl₃) δ -76.1. FAB-HRMS $[M + H]^+$ calcd for $C_{18}H_{24}N_6O_6F_3$ 477.1709, found

5'-O-(4,4'-Dimethoxytrityl)-2'-O-(2-(trifluoroacetamido)ethyl)-N⁶-isobutyryladenosine (A10). 4,4'-Dimethoxytrityl chloride (0.19 g, 0.57 mmol) was added to a solution of A9 (0.249 g, 0.52 mmol) in pyridine (5 mL) cooled to 0 °C. The resulting solution was warmed to room temperature and stirred for 24 h, and the solvents were removed under vacuum. The residual yellow gum was purified via chromatography with the column packed as described for U8. The column was eluted with a gradient of 0-6% MeOH in CH₂Cl₂ to yield 0.292 g (71%) of **A10** as a yellow foam: R_f 0.35 (1:9 MeOH/CH₂Cl₂); ¹H NMR (500 MHz, CDCl₃) δ 8.88 (br s, 1H), 8.60 (s, 1H), 8.22 (s, 1H), 7.37–7.15 (m, 9H), 6.75 (d, $J=8.9~{\rm Hz},\,4{\rm H}$), 6.13 (d, J = 2.6 Hz, 1H), 4.58-4.55 (m, 2H), 4.25 (br q, J = 4.0 Hz, 1H), 3.92 (ABqdd, J = 10.0, 5.0, 3.5 Hz, 1H), 3.81 (ABqdd, $J=11.0,\,7.5,\,3.5~{\rm Hz},\,1{\rm H}),\,3.72~({\rm s},\,6{\rm H}),\,3.54~({\rm m},\,1{\rm H}),\,3.51~({\rm dd},\,1{\rm H}),\,3.51~({\rm dd$ J = 11.0, 3.5 Hz, 1H, 3.48 (m, 1H), 3.37 (dd, J = 10.5, 4.0 Hz,1H), 3.18 (septet, J = 7.5 Hz, 1H), 1.24 (d, J = 6.8 Hz, 6H); $^{13}{\rm C}$ NMR (125 MHz, CDCl₃) δ 176.7, 158.8, 157.8, 152.6, 151.1, 149.6, 144.7, 141.9, 135.9, 135.8, 130.3, 130.2, 128.4, 128.1, 127.2, 123.0, 116.2, 113.4, 88.2, 86.7, 84.0, 82.2, 69.9, 69.0, 63.1, 55.4, 40.2, 36.3, 19.4; ¹⁹F NMR (470 MHz, CDCl₃) δ -76.1. ESI-HRMS $[M + H]^+$ calcd for $C_{39}H_{42}N_6O_8F_3$ 779.3016, found 779.3013.

5'-O-(4,4'-Dimethoxytrityl)-2'-O-(2-(trifluoroacetamido)ethyl)-N⁶-isobutyryladenosine-3'-O-(2-cyanoethyl-N,Ndiisopropyl)phosphoramidite (A11). A portion of A10 (0.214 g, 0.30 mmol) was coevaporated with pyridine (5 mL) and dissolved in CH₂Cl₂ (6 mL). The solution was cooled to 0 °C, and 2-cyanoethyl N,N,N',N'-tetraisopropylphosphorodiamidite (220 µL, 0.69 mmol) and 4,5-dicyanoimidazole (40 mg, 0.34 mmol) were added. The cloudy mixture was warmed to room temperature, stirred for 4 h, and diluted with CH2Cl2 (50 mL) and saturated aqueous NaHCO₃ (50 mL). The aqueous layer was extracted with CH_2Cl_2 (2 × 25 mL). The combined organic extracts were washed with saturated aqueous NaCl (25 mL), dried over Na₂SO₄, and concentrated under vacuum. The residual colorless viscous oil was purified via chromatography on Et₃N-washed silica gel with 1:6:3 to 1:3:6 Et₃N/ hexanes/CH₂Cl₂ as the eluant to yield 0.198 g (73%) of **A11** as a yellow gum. TLC revealed the presence of both diastereomers: R_f 0.48 and 0.38 (1:3:6 Et₃N/hexanes/CH₂Cl₂). NMR spectroscopy also revealed the presence of both diastereomers: ${}^{1}\text{H}$ NMR (500 MHz, acetone- d_{6}) δ 9.31 (br s, 1H), 8.54 and 8.53 (each s, total 1H), 8.46 and 8.45 (each s, total 1H), 7.47-7.21 (m, 9H), 6.87-6.81 (m, 4H), 6.25 and 6.24 (each d, J = 4.5 Hz, total 1H), 5.08-5.05 (m, total 1H), 4.94 and 4.88(each dt, J = 10.5, 5.0 Hz, total 1H), 4.41 and 4.36 (each td, J = 4.5, 3.5 Hz, total 1H), 3.93 (m, total 1H), 3.84 (m, total 1H), 3.78 and 3.77 (each s, total 3H), 3.75–3.73 (m, total 1H), 3.72-3.67 (m, total 2H), 3.55-3.53 (m, total 3H), 3.44-3.40 (m, total 2H), 2.89 and 2.86 (br s, total 1H), 2.82-2.79 (m, total 1H), 2.62 (t, J = 6.0 Hz, total 2H), 1.24–1.12 (m, 18H); ¹³C NMR (125 MHz, acetone- d_6) δ 175.8, 158.97, 158.95, 152.1, 151.6, 150.1, 145.3, 143.1, 142.9, 136.08, 136.02, 135.98, 130.39, 130.32, 130.29, 128.40, 128.33, 127.9, 126.9, 123.5, 118.2, 113.2, 88.1, 87.8, 86.4, 83.5, 80.47, 80.44, 71.63, 71.53, 68.63, 68.49, 63.3, 63.1, 58.82, 58.66, 54.8, 43.33, 43.24, 39.73, $39.61, 35.2, 24.41, 24.35, 24.22, 24.16, 20.04, 19.98, 18.9; {}^{19}F$ NMR (470 MHz, acetone- d_6) δ -76.77 and -76.80; ³¹P NMR (202 MHz, acetone- d_6) δ 150.7 (just one peak observed). FAB-HRMS $[M + Na]^+$ calcd for $C_{48}H_{58}N_8O_9PNaF_3$ 1001.3914, found 1001.3914.

3',5'-Bis-O-(tert-butyldimethylsilyl)-2'-O-(2-(triphenylmethylthio)ethyl)-N⁶-isobutyryladenosine (A12). Diethyl azodicarboxylate (DEAD, 1.13 mL, 7.1 mmol) was added to a solution of triphenylphosphine (1.86 g, 7.1 mmol) in THF (30 mL) cooled to 0 °C. The DEAD was prepared as described;⁴¹ the use of commercially available diisopropyl azodicarboxylate (see procedures for U11 and G12) was not tested. After 1 h, a solution of A5 (2.88 g, 4.72 mmol) and trityl mercaptan (1.95 g, 7.1 mmol) in THF (20 mL) was added through a cannula; the flask containing the latter reagents was rinsed with THF $(6 \times 5 \text{ mL})$. After 1 h, the solution was quenched with MeOH (10 mL) and concentrated under vacuum. The resulting oil was purified via chromatography with 10-40% EtOAc in hexanes as the eluant to produce 3.29 g (80%) of **A12** as an off-white foam: R_f 0.38 (1:2 EtOAc/hexanes); ¹H NMR (500 MHz, CDCl₃) δ 8.68 (s, 1H), 8.63 (br s, 1H), 8.31 (s, 1H), 7.37–7.19 (m, 15H), 6.07 (d, J = 3.8 Hz, 1H), 4.44 (dd, J = 5.1, 4.9 Hz, 1H), 4.18(dd, J = 4.4, 4.1 Hz, 1H), 4.10 (ddd, J = 5.1, 3.4, 2.7 Hz, 1H),3.99 (ABqd, J = 11.5, 3.9 Hz, 1H), 3.77 (ABqd, J = 11.5, 2.7 (ABqd, J = 11Hz, 1H), 3.47 (dt, J = 9.8, 6.8 Hz, 1H), 3.34 (dt, J = 9.9, 6.8Hz, 1H), 3.23 (m, 1H), 2.42 (t, J = 6.9 Hz, 2H), 1.33 (d, J =6.8 Hz, 3H), 1.32 (d, J = 6.8 Hz, 3H), 0.93 (s, 9H), 0.91 (s, 3Hz)9H), 0.11 (s, 6H), 0.10 (s, 3H), 0.06 (s, 3H); ¹³C NMR (125 MHz, $CDCl_3$) δ 176.0, 152.5, 150.8, 149.2, 144.6, 141.6, 129.5, 127.8, 126.6, 122.4, 87.0, 85.0, 82.0, 69.6, 69.4, 66.6, 61.7, 36.1, 31.6, 26.0, 25.7, 19.2, 19.2, 18.4, 18.0, -4.6, -4.9, -5.5. EI-HRMS M^+ calcd for $C_{47}H_{65}N_5O_5SSi_2$ 867.4244, found 867.4242.

2'-O-(2-(Triphenylmethylthio)ethyl)-N⁶-isobutyryl ad**enosine** (A13). TBAF (1 M) in THF (8.4 mL, 8.4 mmol) dropwise over 10 min was added to a solution of A12 (3.29 g, 3.8 mmol) in THF (40 mL) cooled to 0 °C. After 2 h, the solvents were removed under vacuum. The resulting oil was purified via chromatography with 0-5% MeOH in CH₂Cl₂ as the eluant to yield 2.08 g (85%) of **A13** as a white foam: R_f 0.23 (1:19 MeOH/CH₂Cl₂); ¹H NMR (500 MHz, CDCl₃) δ 8.68 (s, 1H), 8.45 (br s, 1H), 7.99 (s, 1H), 7.40-7.21 (m, 15H), 6.11 (dd, J = 12, 12)2 Hz, 1H), 5.84 (d, J = 7.5 Hz, 1H), 4.59 (dd, J = 7.5, 4.4 Hz, 1H), 4.35 (d, J = 3.5 Hz, 1H), 4.34 (br s, 1H), 3.96 (td, J =12.5, 2.5 Hz, 1H), 3.76 (dt, J = 12.5, 1.5 Hz, 1H), 3.26 (septet,J = 6.5 Hz, 1H), 3.19 (dt, J = 10.0, 5.0 Hz, 1H), 3.10 (br s, 1H), 2.86 (ddd, J = 9.7, 8.6, 4.6 Hz, 1H), 2.54 (ddd, J = 13.6, 8.4, 4.9 Hz, 1H), 2.32 (dt, J=13.8, 4.4 Hz, 1H), 1.33 (d, J=13.8, 4.4 Hz, 1H), 1.35 (d, J=13.8, 4.5 Hz, 1H), 1.35 (d, J=13.8, 4.5 Hz, 1H), 1.35 (d, J=13.8, 4.6 Hz, 1H), 1.35 (d, J=13.8, 4.7 Hz, 1H), 1.35 (d, J=13.8, 4.8 Hz, 1H), 1.35 (d, 6.8 Hz, 6H); $^{13}{\rm C}$ NMR (100 MHz, CDCl3) δ 176.3, 151.9, 150.2, 149.9, 144.2, 143.0, 129.3, 128.0, 126.8, 123.4, 89.4, 88.0, 81.0, 70.4, 69.2, 67.0, 63.1, 36.0, 31.8, 19.2, 19.1. FAB-HRMS M+ calcd for C₃₅H₃₈N₅O₅S 640.2593, found 640.2596.

5'-O-(4,4'-Dimethoxytrityl)-2'-O-(2-(triphenylmethylthio)ethyl)- N^6 -isobutyryladenosine (A14). Et₃N (1 mL) and 4,4'-dimethoxytrityl chloride (0.58 g, 1.68 mmol) were added to a solution of A13 (0.98 g, 1.53 mmol) in CH₂Cl₂ (15 mL) cooled to 0 °C. The solution was warmed to room temperature, stirred for 24 h, and quenched with MeOH (2 mL). The sample was purified via chromatography using the dry loading technique (4 g of silica gel) with the column packed as described for U8 (the sample was loaded in 10:9 hexanes/CH₂-Cl₂). The column was eluted with a gradient of Et₃N/hexanes/ CH₂Cl₂ from 1:10:9 to 1:1:18 to produce 0.905 g (63%) of A14 as a yellow foam: R_f 0.17 (1:10:9 Et₃N/hexanes/CH₂Cl₂); ¹H NMR (500 MHz, CDCl₃) δ 8.58 (s, 1H), 8.35 (br s, 1H), 8.10 (s, 1H), 7.42-7.16 (m, 24H), 6.81-6.80 (m, 4H), 6.04 (d, J=4.3Hz, 1H), 4.43 (t, J = 4.5 Hz, 1H), 4.36 (q, J = 5.0 Hz, 1H), 4.20 (q, J = 4.2 Hz, 1H), 3.78 (s, 6H), 3.55 (ABqt, J = 10.5, 5.0)Hz, 1H), 3.49 (ABqd, J = 10.6, 3.5 Hz, 1H), 3.38 (ABqd, J =10.7, 4.4 Hz, 1H), 3.18 (br septet, J = 6.0 Hz, 1H), 3.12 (ABqdd,J = 10.1, 7.9, 5.3 Hz, 1H, 2.87 (br d, J = 5.5 Hz, 1H), 2.57 (m,1H), 2.42 (ABqt, J = 13.5, 5.5 Hz, 1H), 1.31 (d, J = 6.9 Hz, 3H), 1.30 (d, J = 6.9 Hz, 3H); ¹³C NMR (125 MHz, CDCl₃) δ 175.9, 158.6, 152.5, 151.0, 149.2, 144.5, 144.4, 141.5, 135.6, 135.5, 130.0, 129.5, 128.1, 128.0, 127.9, 127.0, 126.8, 122.6,

⁽⁴¹⁾ Simionescu, C. I.; Comanita, E.; Pastravanu, M.; Comanita, B. Acta Chim. Hung. 1990, 127, 41-44.

113.2, 86.8, 86.6, 84.0, 81.4, 69.6, 69.5, 67.0, 62.9, 55.2, 36.1, 31.9, 19.1. FAB-HRMS $[M+Na]^+$ calcd for $C_{56}H_{55}N_5O_7NaS$ 964.3720, found 964.3718.

5'-O-(4,4'-Dimethoxytrityl)-2'-O-(2-(triphenylmethylthio) ethyl)- N^6 -isobutyryladenosine-3'-O-(2-cyanoethyl-N,N-diisopropyl)phosphoramidite (A15). A portion of A14 (0.116 g, 0.12 mmol) was coevaporated with pyridine (3 mL) and dissolved in CH₂Cl₂ (3.4 mL). The solution was cooled to 0 °C, and 2-cyanoethyl N,N,N',N'-tetraisopropylphosphorodiamidite (50 µL, 0.16 mmol) and 4,5-dicyanoimidazole (19 mg, 0.16 mmol) were added. The cloudy mixture was warmed to room temperature, stirred for 4 h, and diluted with CH₂Cl₂ (50 mL) and saturated aqueous NaHCO₃ (50 mL). The aqueous layer was extracted with CH₂Cl₂ (2 × 25 mL). The combined organic extracts were washed with saturated aqueous NaCl (25 mL), dried over Na₂SO₄, and concentrated under vacuum. The crude product was purified via chromatography with the column packed as described for U8 (the sample was loaded in $8:1 \text{ hexanes/CH}_2\text{Cl}_2$). The column was eluted with 1:8:1 to 1:5:4Et₃N/hexanes/CH₂Cl₂ to produce 0.102 g (73%) of **A15** as a white foam: R_f 0.46 (1:5:4 Et₃N/hexanes/CH₂Cl₂). NMR spectroscopy revealed the presence of both diastereomers: ¹H NMR $(400 \text{ MHz}, \text{CDCl}_3) \delta 8.59 \text{ and } 8.56 \text{ (each s, total 1H)}, 8.29 \text{ and}$ 8.28 (each br s, total 1H), 8.14 and 8.09 (each s, total 1H), 7.42-7.15 (m, 24H), 6.81-6.77 (m, 4H), 6.04 and 6.01 (each d, J = 5.6 and 4.8 Hz, total 1H), 4.67 and 4.56 (each t, J = 4.2and 4.8 Hz, total 1H), 4.53-4.43 (m, 1H), 4.36 and 4.30 (each q, J = 4.0 Hz, total 1H), 3.77 and 3.76 (each s, total 6H), 3.70– 3.48 (m, 4H), 3.34-3.21 (m, 3H), 2.57-2.33 (m, 6H), 1.31 (d, J = 6.8 Hz, 6H), 1.15 (d, J = 6.8 Hz, 3H), 1.16 (d, J = 6.8 Hz, 3H), 1.11 (d, J = 6.8 Hz, 3H), 1.02 (d, J = 6.8 Hz, 3H); ¹³C NMR (125 MHz, CDCl₃) δ 175.9, 158.5, 158.5, 152.4, 151.1, $\begin{array}{c} 151.0,\ 149.2,\ 149.2,\ 144.6,\ 144.5,\ 144.5,\ 144.4,\ 144.4,\ 142.0,\\ 141.8,\ 135.6,\ 135.5,\ 130.1,\ 130.0,\ 129.5,\ 129.5,\ 128.3,\ 128.1,\\ \end{array}$ 127.9, 127.8, 126.7, 126.6, 122.6, 117.7, 117.3, 113.1, 87.2, 87.0, 86.6, 86.5, 83.6, 80.7, 80.3, 71.1, 71.0, 70.6, 69.6, 69.3, 66.6, $62.9,\ 62.4,\ 58.8,\ 58.7,\ 57.9,\ 57.8,\ 55.2,\ 55.1,\ 52.6,\ 46.8,\ 43.3,$ 43.2, 43.1, 43.0, 36.0, 31.7, 31.6, 24.7, 24.6, 24.5, 24.5, 20.8, 20.3, 20.2, 20.1, 20.0, 19.0; 31 P NMR (162 MHz, CDCl₃) δ 151.5 and 150.9. FAB-HRMS $[M + H]^+$ calcd for $C_{65}H_{73}N_7O_8PS$ 1142.4979, found 1142.4982

2'-O-(2-Benzyloxyethyl)-2-aminoadenosine (G2). A 60% dispersion of sodium hydride in mineral oil (0.27 g, 6.6 mmol) was added to a suspension of 2,6-diaminopurine riboside (1.56 g, 6.0 mmol) in DMF (20 mL). After the mixture was stirred for 1 h, benzyl 2-bromoethyl ether (0.94 mL, 6.0 mmol) was added dropwise over 5 min, and the pale yellow mixture was stirred for 18 h. The solvents were removed under vacuum, and the residual yellow solid was purified via chromatography with 5-30% MeOH in CH₂Cl₂ as the eluant to produce 0.81 g (33%) of **G2** as a white solid: R_f 0.65 (1:9 MeOH/CH₂Cl₂); ¹H NMR (500 MHz, DMSO- d_6) δ 7.94 (s, 1H), 7.29–7.17 (m, 5H), 6.79 (br s, 2H), 5.84 (d, J = 6.5 Hz, 1H), 5.74 (br s, 2H), 5.45(dd, J = 6.0, 5.0 Hz, 1H), 5.07 (dd, J = 4.5, 1.5 Hz, 1H), 4.47(dd, J = 6.5, 5.5 Hz, 1H), 4.36 (s, 2H), 4.26 (m, 1H), 3.92 (m, 4.26 (m, 1H), 3.92 (m, 4.26 (m, 1H), 4.26 (m, 4H))1H), 3.72 (ABqd, J = 11.1, 14.1 Hz, 1H), 3.60-3.52 (m, 3H), 3.49 (t, 4.5 Hz, 2H); 13 C NMR (125 MHz, DMSO- d_6) δ 160.9, 156.9, 152.2, 138.9, 136.7, 128.9, 128.2, 128.0, 114.1, 86.8, 85.7, 81.6, 72.0, 69.8, 69.7, 69.5, 62.3. FAB-HRMS $[M + H]^+$ calcd for C₁₉H₂₅N₆O₅ 417.1886, found 417.1888.

2'-O-(2-Benzyloxyethyl)guanosine (G3). A solution of adenosine deaminase type II (20 mg, 30 units) in 0.1 M Tris, pH 7.5 (4 mL), 0.1 M sodium phosphate, pH 7.4 (0.2 mL), and DMSO (1.0 mL) was added to a solution of G2 (0.65 g, 1.6 mmol) in 0.1 M Tris, pH 7.5 (6 mL), 0.1 M sodium phosphate, pH 7.4 (0.4 mL), and DMSO (1.5 mL). The cloudy white mixture was stirred at room temperature for 36 h, and the solvents were removed under vacuum. The pale yellow viscous oil was purified via chromatography with 5–15% MeOH in CH₂Cl₂ as the eluant to yield 0.53 g (82%) of G3 as white powder: R_f 0.35 (1:9 MeOH/CH₂Cl₂); ¹H NMR (400 MHz, DMSO- d_6) δ 10.62 (br s, 1H), 7.95 (s, 1H), 7.27–7.17 (m, 5 H),

6.45 (br s, 2H), 5.80 (d, J=6.4 Hz, 1H), 5.08–5.04 (m, 2H), 4.37 (s, 1H), 4.35 (m, 1H), 4.23 (dd, J=4.5, 3.0 Hz, 1H), 3.88 (q, J=3.2 Hz, 1H), 3.72 (m, 1H), 3.62–3.55 (m, 2H), 3.55–3.44 (m, 4H); $^{13}\mathrm{C}$ NMR (100 MHz, DMSO- d_6) δ 156.9, 154.2, 151.5, 138.2, 135.6, 128.3, 127.6, 127.5, 116.7, 86.0, 84.6, 81.5, 72.1, 69.2, 69.1, 69.0, 61.5. ESI-HRMS [M + H] $^+$ calcd for $\mathrm{C}_{19}\mathrm{H}_{24}\mathrm{N}_5\mathrm{O}_6$ 418.1727, found 418.1735.

 $3',5',O^6$ -Tris(tert-butyldimethylsilyl)-2'-O-(2-benzyloxyethyl)guanosine (G4). tert-Butyldimethylsilyl chloride (0.72 g, 4.8 mmol) and imidazole (0.55 g, 7.9 mmol) were added to a solution of G3 (0.53 g, 1.3 mmol) in DMF (15 mL). The solution was stirred at room temperature for 12 h, diluted with CH₂Cl₂ (15 mL), and quenched with MeOH (1 mL) and water (20 mL). The aqueous layer was extracted with CH_2Cl_2 (3 \times 20 mL). The combined organic extracts were washed with saturated aqueous NH₄Cl (20 mL) and saturated aqueous NaCl (20 mL) and dried over MgSO₄. The solvents were removed under vacuum to produce 0.78 g (73%) of G4 as a white powder: R_f 0.55 (1:9 MeOH/CH₂Cl₂); ¹H NMR (400 MHz, CDCl₃) δ 7.83 (s, 1H), 7.30–7.24 (m, 5H), 6.36 (br s, 2H), 5.97 $(\mathrm{d}, J = 4.0 \; \mathrm{Hz}, \, 1\mathrm{H}), \, 4.52 \; (\mathrm{s}, \, 2\mathrm{H}), \, 4.48 \; (\mathrm{t}, \, J = 4.8 \; \mathrm{Hz}, \, 1\mathrm{H}), \, 4.24$ (t, J = 4.0 Hz, 1H), 4.08 (dt, J = 5.6, 2.9 Hz, 1H), 3.93 (ABqd,J = 11.4, 3.3 Hz, 1H, 3.81 - 3.74 (m, 3H), 3.61 (t, J = 4.0 Hz,2H), 0.92 (s, 18H), 0.91 (s, 9H), 0.11 (s, 3H), 0.10 (s, 6H), 0.09(s, 3H), 0.08 (s, 6H); 13 C NMR (100 MHz, CDCl₃) δ 159.4, 153.7, $151.5,\, 138.3,\, 136.2,\, 128.6,\, 127.9,\, 127.8,\, 117.7,\, 86.7,\, 84.8,\, 82.8,\, 127.9,\, 127.8,\, 117.7,\, 127.9,\, 127.8,\, 117.7,\, 127.9,\, 127.9,\, 127.8,\, 117.7,\, 127.9,\,$ 73.4, 70.4, 70.2, 69.6, 62.1, 26.3, 26.0, 25.9, 18.7, 18.4, -4.3,-4.6, -5.1, -5.2. ESI-HRMS [M + H]⁺ calcd for $C_{37}H_{66}N_5O_{6-}$ Si₃ 760.4321, found 760.4298.

3',5'-Bis(tert-butyldimethylsilyl)-2'-O-(2-benzyloxyethyl)-N²-isobutyrylguanosine (G5). Isobutyryl chloride (0.50 mL, 5.0 mmol) was added to a solution of G4 (0.78 g, 1.03 mmol) in pyridine (5 mL) cooled to 0 °C. After the mixture was stirred for 45 min, the colorless solution was quenched with saturated aqueous NaHCO₃ (25 mL) and extracted with EtOAc (2 \times 25 mL). The combined organic extracts were dried over Na₂SO₄ and concentrated under vacuum. The residual yellow foam was purified via chromatography with 1-5% MeOH in CH₂Cl₂ as the eluant to yield $0.\overline{73}$ g (98%) of **G5** as a white foam: R_f 0.75 (1:19 MeOH/CH₂Cl₂); ¹H NMR (500 MHz, CDCl₃) δ 8.92 (br s, 1H), 8.05 (s, 1H), 7.29 (s, 1H), 7.28-7.24 (m, 5H), 5.98 (d, J = 4.0 Hz, 1H), 4.52 (ABq, J = 12.4 Hz, 1H), 4.43 (ABq)J = 12.0 Hz, 1H), 4.42 (d, J = 3.0 Hz, 1H), 4.10 (t, J = 4.0 Hz, 1H)1H), 4.07 (dt, J = 5.5, 3.0 Hz, 1H), 3.91 (ABqd, J = 11.6, 2.8 Hz, 1H), 3.76 (ABqd, $J=11.6,\,3.0$ Hz, 1H), 3.74 (br q, J=3.0Hz, 2H), 3.55 (t, J = 5.0 Hz, 2H), 2.52 (septet, J = 7.0 Hz, 1H), 1.16 (d, J = 7.0 Hz, 3H), 1.14 (d, J = 6.5 Hz, 3H), 0.91 (s, J = 6.5 Hz, 3H)9H), 0.90 (s, 9H), 0.10-0.09 (m, 12H); ¹³C NMR (125 MHz, $CDCl_3$) δ 178.9, 155.9, 148.2, 147.8, 138.2, 137.2, 128.6, 127.9, 127.8, 121.6, 86.5, 85.3, 83.6, 73.5, 70.1, 69.7, 62.1, 36.5, 25.9, 25.6, 18.8, 18.8, 18.3, 18.0, -4.7, -4.9, -5.4, -5.5. ESI-HRMS $[M + H]^+$ calcd for $C_{35}H_{58}N_5O_7Si_2$ 716.3875, found 716.3860.

3',5'-Bis(tert-butyldimethylsilyl)-2'-O-(2-hydroxyethyl)- N^2 -isobutyrylguanosine (G6). Pd/C (10%, 1.0 g, 0.94 mmol) was added to a solution of G5 (0.73 g, 1.02 mmol) in EtOH (10 mL). The flask was evacuated and placed under an atmosphere of hydrogen. The consumption of G5 was monitored by TLC. After 12 h, 10 g of Celite was added, and the mixture was filtered and concentrated under vacuum. The residue was purified via chromatography with 0-5% MeOH in CH_2Cl_2 as the eluant to produce 0.56 g (88%) of **G6** as white foam: R_f 0.45 (1:19 MeOH/CH₂Cl₂); ¹H NMR (500 MHz, CDCl₃) δ 10.37 (br s, 1H), 8.04 (s, 1H), 6.01 (s, 1H), 4.38 (t, J = 5.5Hz, 1H), 4.07 (d, J = 7.0 Hz, 1H), 4.05 (d, J = 5.0 Hz, 1H), 3.91 (q, J = 5.07 Hz, 1H), 3.75 (ABq, J = 5.0 Hz, 1H), 3.793.71 (br m, 6H), 2.75 (septet, J = 6.5 Hz, 1H), 1.18 (d, J = 7.0Hz, 6H), 0.87 (s, 9H), 0.86 (s, 9H), 0.06 (s, 6H), 0.05 (s, 3H), 0.04 (s, 3H); $^{13}{\rm C}$ NMR (125 MHz, CDCl3) δ 180.1, 156.3, 148.4, 147.9, 137.1, 121.3, 88.2, 84.4, 84.2, 72.7, 69.3, 61.7, 61.2, 36.3, 26.3, 25.6, 19.3, 19.2, 19.1, 19.0, 18.6, 18.2, -4.4, -4.8, -5.2,-5.3. ESI-HRMS [M + H]⁺ calcd for $C_{28}H_{52}N_5O_7Si_2$ 626.3405, found 626.3394.

3',5'-Bis(tert-butyldimethylsilyl)-2'-O-(2-methanesulfonylethyl)-N2-isobutyrylguanosine (G7). A solution of G6 (0.50 g, 0.80 mmol) in CH₂Cl₂ (10 mL) was cooled to 0 °C. Portions of Et₃N (0.15 mL, 1.07 mmol) and methanesulfonyl chloride (84 μ L, 1.09 mmol) were added. After 1 h, the colorless solution was diluted with CH2Cl2 (25 mL), washed with saturated aqueous NaHCO₃ (2 × 25 mL), dried over Na₂SO₄, and concentrated under vacuum to produce 0.55 g (98%) of **G7** as a pale yellow foam: R_f 0.45 (1:24 MeOH/CH₂Cl₂); ¹H NMR (400 MHz, CDCl₃) δ 12.18 (br s, 1H), 9.90 (br s, 1H), 8.06 (s, 1H), 5.90 (s, 1H), 4.38-4.31 (m, 4H), 3.99 (t, J = 8.8 (s, 1H), 3.99 (t, 2H), 3.99Hz, 1H), 3.94 (m, 1H), 3.90 (d, J = 4.8 Hz, 1H), 3.73 (d, J = 4.8 Hz, 1H), 3.74 (d, J = 4.8 Hz, 1H), 3.74 (d, J = 4.8 Hz, 1H), 3.74 (d, J = 4.8 Hz, 1H), 3.75 (11.2 Hz, 1H), 3.07 (s, 3H), 2.67 (septet, J = 6.8 Hz, 1H), 1.14 (d, J = 6.8 Hz, 3H), 1.12 (d, J = 6.8 Hz, 3H), 0.87 (s, 9H), 0.82(s, 9H), 0.08 (s, 3H), 0.06 (s, 3H), 0.02 (s, 3H), 0.01 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 180.2, 156.0, 148.4, 147.8, 136.8, 121.6, 88.1, 84.4, 83.7, 69.7, 69.5, 68.8, 60.9, 51.0, 37.7, 36.3, 26.3, 25.8, 19.2, 19.1, 18.7, 18.2, -4.4, -4.7-5.1, -5.2. ESI-HRMS $[M + H]^+$ calcd for $C_{29}H_{54}N_5O_9Si_2S$ 704.3181, found 704.3171.

3',5'-Bis(tert-butyldimethylsilyl)-2'-O-(2-azidoethyl)- N^2 -isobutyrylguanosine (G8). NaN₃ (0.25 g, 4.0 mmol) and 18-crown-6 (50 mg, 0.19 mmol) were added to a solution of G7 (0.55 g, 0.78 mmol) in DMF (6 mL). The cloudy pale yellow mixture was stirred at 60 °C for 5 h, diluted with EtOAc (25 mL), washed with saturated aqueous NaHCO₃ (2 × 25 mL), dried over Na₂SO₄, and concentrated under vacuum. The pale yellow foam was purified via chromatography using a gradient of 20-40% EtOAc in CH₂Cl₂ as the eluant to produce 0.35 g (70%) of **G8** as a white foam: R_f 0.75 (1:2 EtOAc/CH₂Cl₂); ¹H NMR (400 MHz, CDCl₃) δ 12.37 (s, 1H), 10.35 (s, 1H), 8.14 (s, 1H), 5.98 (d, J = 2.4 Hz, 1H), 4.39 (dd, J = 7.4, 5.2 Hz, 1H), 4.05 (td, J = 9.2, 2.4 Hz, 1H), 3.98 (dd, J = 12.0, 2.4 Hz, 1H),3.89 (dt, J = 9.2, 2.4 Hz, 1H), 3.99 (ABqd, J = 11.6, 2.4 Hz, 1H), 3.85 (dd, J = 4.4, 2.0 Hz, 1H), 3.85 (m, 1H), 3.75 (ABqd, J = 11.2, 2.4 Hz, 1H, 3.71 (m, 1H), 2.76 (septet, J = 6.8 Hz,1H), 1.15 (d, J = 6.8 Hz, 3H), 1.12 (d, J = 6.8 Hz, 3H), 0.87 (s, 9H), 0.85 (s, 9H), 0.08 (s, 3H), 0.07 (s, 3H), 0.031 (s, 3H), 0.028 (s, 3H); $^{13}\mathrm{C}$ NMR (100 MHz, CDCl₃) δ 180.2, 156.2, 148.4, 148.3, 137.3, 121.4, 87.0, 84.1, 83.7, 69.7, 69.0, 68.8, 61.3, 51.0, 36.2, 26.3, 25.8, 19.2, 18.7, 18.2, -4.4, -4.7, -5.2. ESI-HRMS $[M + H]^+$ calcd for $C_{28}H_{51}N_8O_6Si_2$ 651.3470, found 651.3467.

2'-O-(2-(Trifluoroacetamido)ethyl)- N^2 -isobutyrylguanosine (G9). Pd/C (10%, 0.37 g, 0.35 mmol) was added to a solution of G8 (0.350 g, 0.53 mmol) in EtOH (13 mL). The flask was evacuated and placed under an atmosphere of hydrogen. The consumption of G8 was monitored by TLC. After 12 h, 10 g of Celite was added, and the mixture was filtered and concentrated under vacuum. The pale vellow foam was dissolved in THF (5 mL), and ethyl trifluoroacetate (0.28 mL, 2.4 mmol) and Et₃N (0.34 mL, 2.4 mmol) were added. After 12 h, the solution was cooled to 0 °C, and 1 M TBAF in THF (1.24 mL, 1.24 mmol) was added dropwise over 5 min. The solution was warmed to room temperature and stirred for 6 h. The solvents were removed under vacuum, and the residual yellow foam was purified via chromatography using a gradient of 5-15% MeOH in CH₂Cl₂ as the eluant to yield 0.190 g (73%) of **G9** as a white foam: R_f 0.23 (1:19 MeOH/CH₂Cl₂); ¹H NMR (400 MHz, DMSO- d_6) δ 12.06 (br s, 1H), 11.62 (br s, 1H), 9.25 (s, 1H), 5.88 (d, J = 6.0 Hz, 1H), 5.26 (d, J = 4.8 Hz, 1H), 5.07(t, J = 5.2 Hz, 1H), 4.38 (t, J = 4.8 Hz, 1H), 4.30 (ABq, J = 4.8 Hz, 1H)4.8 Hz, 1H), 3.92 (ABq, J = 4.8 Hz, 1H), 3.67 (m, 1H), 3.55(m, 4H), 2.74 (septet, $\tilde{J} = 6.8$ Hz, 1H), 1.11 (s, 3H), 1.09 (s, 3H); 13 C NMR (100 MHz, DMSO- d_6) δ 180.8, 155.5, 149.5, 148.9, 138.1, 120.8, 86.7, 85.2, 82.2, 69.5, 67.9, 61.8, 35.4, 19.49, 19.48l; ¹⁹F NMR (376 MHz, DMSO- d_6) δ -74.9. ESI-HRMS $[M + H]^+$ calcd for $C_{18}H_{24}N_6O_7F_3$ 493.1659, found 493.1664.

5′-O-(4,4′-Dimethoxytrityl)-2′-O-(2-(trifluoroacetamido)-ethyl)-N²-isobutyrylguanosine (G10). A portion of G9 (0.180 g, 0.37 mmol) was coevaporated from pyridine (2 × 5 mL) and dissolved in pyridine (3 mL). The solution was cooled to 0 °C, and 4,4′-dimethoxytrityl chloride (0.15 g, 0.44 mmol)

was added. After the solution was stirred at room temperature for 18 h, the solvents were removed under vacuum. The residual vellow foam was purified via chromatography with the column packed as described for **U9**. The column was eluted with a gradient of 0-5% MeOH in CH₂Cl₂ to produce 0.240 g (83%) of **G10** as a white foam: R_f 0.45 (1:19 MeOH/CH₂Cl₂); ¹H NMR (500 MHz, acetone- d_6) δ 10.68 (br s, 1H), 8.91 (br s, 1H), 8.01 (s, 1H), 7.45-7.17 (m, 9H), 6.81 (m, 4H), 6.03 (d, J = 3.0 Hz, 1H), 4.70 (t, J = 11.0 Hz, 2H), 4.61 (t, J = 7.5 Hz,1H), 4.21 (m, 1H), 3.98 (m, 2H), 3.74(s, 6H), 3.60 (m, 2H), 3.46-3.37 (m, 2H), 3.20 (q, J = 7.0 Hz, 1H), 2.88 (septet, J = 7.0 Hz)Hz, 1H), 1.22 (d, J = 4.0 Hz, 3H), 1.20 (d, J = 3.5 Hz, 3H); ¹³C NMR (125 MHz, acetone- d_6) δ 180.3, 159.0, 158.9, 155.4, 148.7, 148.5, 145.4, 137.7, 136.1, 136.0, 130.3, 128.4, 128.0, 127.0, 121.3, 113.2, 87.3, 86.3, 83.8, 82.3, 70.1, 68.8, 63.9, 54.8, 46.4, 40.0, 35.9, 29.7, 18.7, 18.6; 19 F NMR (470 MHz, acetone- d_6) δ -76.7. ESI-HRMS $[M + H]^+$ calcd for $C_{39}H_{42}N_6O_9F_3$ 795.2965, found 795.2978.

5'-O-(4,4'-Dimethoxytrityl)-2'-O-(2-(trifluoroacetamido)ethyl)- N^2 -isobutyrylguanosine-3'-O-(2-cyanoethyl-N,Ndiisopropyl)phosphoramidite (G11). A portion of G10 (0.160 g, 0.20 mmol) was coevaporated with pyridine (2×5) mL) and dissolved in CH₂Cl₂ (2 mL). The solution was cooled to 0 °C, and 2-cyanoethyl N,N,N',N'-tetraisopropylphosphorodiamidite (87 μ L, 0.27 mmol) and 4,5-dicyanoimidazole (27 mg, 0.24 mmol) were added. The colorless solution was warmed to room temperature, stirred for 6 h, diluted with 5% Et₃N in EtOAc (25 mL), and washed with saturated aqueous NaHCO₃ $(2 \times 25 \text{ mL})$. The organic layer was dried over Na₂SO₄ and concentrated under vacuum. The residual pale yellow solid was purified by chromatography with the column packed as described for **U9** (the sample was loaded in 4:15 hexanes/CH₂-Cl₂). The column was eluted with a gradient from 1:4:15 Et₃N/ hexanes/CH₂Cl₂ to 1:19 Et₃N/CH₂Cl₂ to produce 0.160 g (80%) of **G11** as a white foam: R_f 0.46 (1:19 Et₃N/CH₂Cl₂). NMR spectroscopy revealed the presence of both diastereomers: ¹H NMR (500 MHz, acetone- d_6) δ 8.87 (br s, 1H), 7.99 and 7.94 (each s, total 1H), 7.50-7.23 (m, 9H), 6.87 (m, 4H), 6.01 and 6.00 (each d, J = 4.5 Hz, total 1H), 4.73 (dt, J = 8.5 Hz, 5.0 Hz, 1H), 4.66 and 4.59 (each dt, J = 9.5 Hz, 5.5 Hz, total 1H), 4.38 and 4.37 (each m, total 1H), 4.04 (m, 1H), 3.98 (m, 2H), 3.90 (m, 1H), 3.78 and 3.77 (each s, total 6H), 3.74–3.40 (m, 8H), 2.86 (septet, J = 7.0 Hz, 1H), 2.79 (t, J = 10.5 Hz, 1H), 2.59 (m, 1H), 1.18 (m, 16H), 1.08 (s, 1H), 1.07 (s, 1H); ¹³C NMR $(125 \text{ MHz}, \text{acetone-}d_6) \delta 180.2, 159.0, 155.2, 148.7, 148.5, 145.3,$ 145.2, 136.9, 136.8, 136.0, 135.9, 130.3, 128.4, 128.3, 127.0, 121.6, 118.6, 118.3, 113.3, 87.5, 86.5, 83.5, 83.2, 81.5, 81.2, 71.4, 71.3, 68.6, 68.4, 63.5, 63.2, 59.1, 59.0, 58.5, 58.4, 54.9, 52.8, 43.3, 43.2, 39.8, 35.8, 24.5, 20.0, 18.6; ¹⁹F NMR (470 MHz, acetone- $d_6)\ \delta$ -76.7 (just one peak observed); $^{31}\mathrm{P}\ \mathrm{NMR}$ (202 MHz, acetone- d_6) δ 151.2 and 150.6. ESI-HRMS [M + H]⁺ calcd for C₄₈H₅₉N₈O₁₀F₃P 995.4044, found 995.4040.

2'-O-(2-(Triphenylmethylthio)ethyl)-N²-isobutyrylgua**nosine** (G12). Diisopropyl azodicarboxylate (123 μ L, 0.64 mmol) was added to a solution of triphenylphosphine (0.124 g, 0.50 mmol) in THF (4 mL) cooled to 0 °C. After 30 min, a solution of G6 (0.153 g, 0.24 mmol) and trityl mercaptan (0.178 g, 0.62 mmol) in THF (2 mL) was added via cannula; the flask containing the latter reagents was rinsed with THF (2 \times 2 mL). After 30 min, 1 M TBAF in THF (0.48 mL, 0.48 mmol) was added to the solution. After 3 h, the solvents were removed under vacuum, and the residue was purified via chromatography with 0-2% MeOH in CH₂Cl₂ as the eluant to yield 0.156 g (97%) of **G12** as a white foam: R_f 0.33 (1:19 MeOH/CH₂Cl₂); ${}^{\bar{1}}H$ NMR (500 MHz, CDCl₃) δ 9.69 (br s, 1H), 8.12 (s, 1H), 7.31– 7.13 (m, 15H), 5.86 (d, $J=6.0~{\rm Hz},~1{\rm H}),$ 5.23 (br s, 1H), 4.43 (br s, 1H), 4.39 (q, J = 5.0 Hz, 1H), 4.17 (br s, 1H), 3.95 (br s, 1H)1H), 3.88 (ABq, J = 12.5 Hz, 1H), 3.74 (br ABqd, J = 12.0, 1.5Hz, 1H), 3.19 (m, 3H), 2.77 (septet, 1H, J = 7.0 Hz), 2.41 (ABqt, J = 13.5, 6.0 Hz, 1H), 2.32 (ABqt, J = 13.5, 5.5 Hz, 1H), 1.21 $(d, J = 6.5 \text{ Hz}, 3\text{H}), 1.20 (d, J = 6.9 \text{ Hz}, 3\text{H}); {}^{13}\text{C NMR} (125)$ MHz, CDCl₃) δ 178.9, 155.1, 147.6, 147.2, 144.2, 139.1, 129.3,

127.9, 126.8, 122.0, 88.1, 86.8, 81.7, 70.2, 69.2, 66.5, 62.5, 36.2, 31.9, 18.9. ESI-HRMS $[M\ +\ H]^+$ calcd for $C_{35}H_{38}N_5O_6S$ 656.2543, found 656.2543.

5'-O-(4,4'-Dimethoxytrityl)-2'-O-(2-(triphenylmethylthio)ethyl)- N^2 -isobutyrylguanosine (G13). A portion of G12 (0.150 g, 0.23 mmol) was coevaporated from pyridine (3 mL) and dissolved in pyridine (3 mL). The solution was cooled to 0 °C, and 4,4'-dimethoxytrityl chloride (0.089 g, 0.26 mmol) was added. After the mixture was stirred at room temperature for 12 h, the solvents were removed under vacuum. The residual pale yellow foam was purified via chromatography with the column packed as described for **U9**. The column was eluted with a gradient of 0-5% MeOH in CH₂Cl₂ to yield 0.150 g (68%) of **G13** as a white foam: R_f 0.55 (1:19 MeOH/CH₂Cl₂); ¹H NMR (500 MHz, CDCl₃) δ 7.80 (s, 1H), 7.46 (dd, J = 7.0, 1.5 Hz, 2H), 7.36-7.11 (m, 24H), 6.82-6.75 (m, 4H), 5.83 (d, J = 6.5 Hz, 1H), 4.54 (m, 1H), 4.34 (m, 1H), 4.15 (d, J = 2.0 (m, 1H), 4.10 (m, 1H), 4.10Hz, 1H), 3.75 (s, 3H), 3.74 (s, 3H), 3.40 (ABqd, J = 10.5, 2.0Hz, 1H), 3.39 (ABqt, J = 10.5, 5.0 Hz, 1H), 3.15 (ABqd, J =10.5, 3.5 Hz, 1H), 3.07-3.02 (m, 1H), 2.61 (m, 2H), 2.37 (ABqt, J = 13.0, 5.5 Hz, 1H), 1.91 (septet, J = 7.0 Hz, 1H), 0.91 (d, $J = 7.0 \text{ Hz}, 3\text{H}), 0.81 \text{ (d}, J = 7.0 \text{ Hz}, 3\text{H}); {}^{13}\text{C NMR} (125 \text{ MHz}, 32 \text{ Hz})$ $CDCl_3$) δ 179.1, 158.9, 156.7, 148.5, 147.5, 145.1, 144.6, 138.7, 136.1, 135.8, 130.3, 130.2, 129.7, 128.2, 127.3, 127.1, 122.2, 113.5, 86.5, 84.5, 69.7, 67.5, 55.5, 36.4, 32.1, 18.9. ESI-HRMS $[M + H]^+$ calcd for $C_{56}H_{56}N_5O_8S$ 958.3850, found 958.3817.

5'-O-(4,4'-Dimethoxytrityl)-2'-O-(2-(triphenylmethylthio)ethyl)-N²-isobutyrylguanosine-3'-O-(2-cyanoethyl-N,N-diisopropyl)phosphoramidite (G14). A portion of G13 (0.150 g, 0.16 mmol) was coevaporated with pyridine (2×5) mL) and dissolved in CH₂Cl₂ (1 mL). The solution was cooled to 0 °C, and 2-cyanoethyl N,N,N',N'-tetraisopropylphosphorodiamidite (65 μ L, 0.21 mmol) and 4,5-dicyanoimidazole (24 mg, 0.22 mmol) were added. The colorless solution was warmed to room temperature, stirred for 6 h, diluted with 5% Et₃N in EtOAc (25 mL), and washed with saturated aqueous NaHCO₃ $(2 \times 25 \text{ mL})$. The organic layer was dried over Na₂SO₄ and concentrated under vacuum. The residual pale yellow solid was purified via chromatography with the column packed as described for **U9** (the sample was loaded in 14:6 hexanes/CH₂-Cl2). The column was eluted with a gradient from 1:14:6 to 1:3:16 Et₃N/hexanes/CH₂Cl₂ to produce 0.109 g (61%) of G14 as a white foam: R_f 0.44 (1:3:16 Et₃N/hexanes/CH₂Cl₂). NMR spectroscopy revealed the presence of both diastereomers: ¹H NMR (500 MHz, acetone- d_6) δ 7.98 and 7.96 (each s, total 1H), 7.50 and 7.49 (each s, 2H), 7.36-7.23 (m, 24H), 6.87 (m, 4H), 5.96 and 5.92 (each d, J = 6.0 Hz, total 1H), 4.63 and 4.57(each t, J = 5.5 Hz, total 1H), 4.53 (m, 1H), 4.35 and 4.28 (each ABqd, J = 7.5, 4.0 Hz, total 1H), 3.78 (s, 6H), 3.72–3.59 (m, 5H), 3.45-3.32 (m, 3H), 2.77 (septet, J = 7.0 Hz, 1H), 2.72(m, 1H), 2.60 (t, J = 6.0 Hz, 1H), 2.36 (m, 2H), 1.16 (m, 16H),1.07 (d, J = 7.0 Hz, 2H); ¹³C NMR (125 MHz, acetone- d_6) δ 180.2, 159.0, 155.2, 148.7, 148.5, 145.3, 145.2, 136.9, 136.8, 136.0, 135.9, 130.3, 128.4, 128.3, 127.0, 121.6, 118.6, 118.3, 113.3, 87.5, 86.5, 83.5, 83.2, 81.5, 81.2, 71.4, 71.3, 68.6, 68.4, 63.5, 63.2, 59.1, 59.0, 58.5, 58.4, 54.9, 52.8, 46.1, 43.3, 43.2, 39.8, 35.8, 24.5, 20.0, 18.6; $^{\rm 31}{\rm P}$ NMR (202 MHz, acetone- d_6) δ 151.13 and 151.06. FAB-HRMS $[M + H]^+$ calcd for $C_{65}H_{73}N_7O_{9}$ -PS 1158.4928, found 1158.4929.

Solid-Phase RNA Oligonucleotide Synthesis. Acetonitrile (synthesis grade), the standard 2'-TOM-phosphoramidetes for A, C, G, and U, 5-ethylthio-1H-tetrazole, and other standard solutions were from a commercial supplier. RNA oligonucleotides were synthesized on the 1 μ mol scale. The standard coupling time of 6 min was used for the four standard phosphoramidites, and an increased coupling time of 10 min was used for the modified phosphoramidites. After the solid-phase synthesis, the solid support was transferred to a screw-cap glass vial and incubated at room temperature for 16 h with 1.5 mL of methylamine solution, prepared by mixing equal volumes of 40% aqueous methylamine and 33% methylamine in ethanol. After the vial was cooled briefly on ice, the

supernatant was transferred by pipet into two 1.5 mL eppendorf tubes; the solid support and vial were rinsed with water $(2 \times 0.25 \text{ mL})$. The combined solutions were evaporated to dryness using an evaporating centrifuge. The residue was dissolved in a total volume of 1.0 mL of 1 M TBAF in THF and stirred at 37 °C for 12 h. One milliliter of 1 M Tris, pH 7.5, was added to the solution, and the solution was evaporated to half the original volume. Additional 1 M Tris, pH 7.5, was added to a total volume of 2.0 mL, and the oligonucleotide was desalted on a NAP-25 column using water as the eluant. The fractions containing the desired RNA oligonucleotide (as judged by UV absorbance at 260 nm) were combined and evaporated to dryness. The oligonucleotide was then purified by 20% denaturing PAGE using 1× TBE (89 mM each Tris and boric acid, pH $\bar{8}.3$, containing 2 mM EDTA) as the running buffer. The gel electrophoresis and oligonucleotide extraction and precipitation procedures have been described previously. 46 The solid-phase coupling efficiency for each of the modified phosphoramidites was >90% as judged by the appearance of abort bands on the polyacrylamide gels. The yield of the purified oligoribonucleotide was typically 150-200 nmol, which is similar to the yields for unmodified oligoribonucleotides that are synthesized in a similar fashion.

Oligonucleotide Sequences and Mass Spectrometry Data. Five RNA oligonucleotides were prepared using the above procedure incorporating phosphoramidites U9, U10, A11, G11, and A15. The first four oligonucleotides have one of the four 2'-tethered amine nucleotides (U, C, A, or G), and the fifth oligonucleotide has the adenosine 2'-tethered thiol nucleotide with the thiol group protected as the S-trityl derivative. The oligonucleotide sequences and mass spectrometry data are given below. 5'-32P radiolabeling of each oligonucleotide after PAGE purification revealed that each oligonucleotide is >99% pure (see Supporting Information). The oligonucleotide with the 2'-tethered thiol was ligated to a larger strand of RNA with T4 DNA ligase and a DNA splint, 40 and the S-trityl group was subsequently deprotected with AgNO3 essentially as described 31 (data not shown).

Oligonucleotide with 2'-Tethered Amine U. The sequence is 5'-UUCUGUUGAXAUGGAUGCAGUUCA-3', where X denotes the modified nucleotide from incorporation of phosphoramidite U9. MALDI-MS calcd 7675.6, found 7677.2.

Oligonucleotide with 2'-Tethered Amine C. The sequence is 5'-UUCUGUUGAUAUGGAUGXAGUUCA-3', where X denotes the modified nucleotide from incorporation of phosphoramidite U10. MALDI-MS calcd 7675.6, found 7672.9.

Oligonucleotide with 2'-Tethered Amine A. The sequence is 5'-GGAAUUGCGGGAXAG-3', where X denotes the modified nucleotide from incorporation of phosphoramidite A11. MALDI-MS calcd 4961.1, found 4962.1.

Oligonucleotide with 2'-Tethered Amine G. The sequence is 5'-GGAAUUXCGGGAAAG-3', where X denotes the modified nucleotide from incorporation of phosphoramidite G11. MALDI-MS calcd 4961.1, found 4961.9.

Oligonucleotide with 2'-Tethered Thiol A. The sequence is 5'-GGAAUZGCGGGAXAG-3', where X denotes the modified nucleotide (S-trityl protecting group remaining) from incorporation of phosphoramidite A15 and Z denotes a 2'-amino-2'-deoxyuridine nucleotide (C. V. Miduturu and S. K. Silverman, manuscript in preparation). MALDI-MS calcd 5219.3, found 5218.7.

Acknowledgment. This research was supported by the Burroughs Wellcome Fund (New Investigator Award in the Basic Pharmacological Sciences), the March of Dimes Birth Defects Foundation (Research Grant 5-FY02-271), the National Institutes of Health (GM-65966), the American Chemical Society Petroleum Research Fund (38803-G4), and the UIUC Department of Chemistry (all to S.K.S.). S.K.S. is the recipient of a fellowship from The David and Lucile Packard Founda-

tion. We are grateful to Rebecca Coppins for the initial investigations into 2'-alkylations of uridine nucleosides and to Mary Smalley for the PAGE purification of the RNA oligonucleotides prepared using phosphoramidites U10 and A11. We thank members of the Silverman lab for discussions.

Supporting Information Available: General experimental procedures, $^1\mathrm{H}$ and $^{31}\mathrm{P}$ NMR spectra, and $5'-^{32}\mathrm{P}$ radiolabeling to demonstrate oligoribonucleotide purity. This material is available free of charge via the Internet at http://pubs.acs.org.

JO050061L